



Questions:

- 1) What is happening to the average kinetic energy of the particles in the sample during section 2? How do you know?
- 2) As a substance goes through section (2), what happens to the distance between the particles? How do you know?
- 3) What is the name of the process happening during section (4)?
- 4) What would be the name of the process happening during section (4) if time were going the other way?
- 5) What is the melting point of this substance?
- 6) At what temperature would this sample finish boiling? Explain.
- 7) For section 3, the temperature does not remain constant because:
  - a. Heat is not being absorbed
  - b. The ice is colder than the water
  - c. Heat energy is being converted to potential energy
  - d. Heat energy is being converted to kinetic energy

# Heating Of The Atmosphere Worksheet Answers

**Léo Pomar**



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