

Limiting Reagents and Percentage Yield Worksheet

1. Consider the reaction



- a) 80.0 grams of iodine(V) oxide, I_2O_5 , reacts with 28.0 grams of carbon monoxide, CO . Determine the mass of iodine I_2 , which could be produced?

80 g I_2O_5	1 mol I_2O_5	1 mol I_2		XS
1	333.8 g I_2O_5	1 mol I_2O_5		
28 g CO	1 mol CO	1 mol I_2	253.8 g I_2	
1	28 g CO	5 mol CO	1 mol I_2	= 50.8 g I_2

- b) If, in the above situation, only 0.160 moles, of iodine, I_2 was produced.
i) what mass of iodine was produced?

0.160 mol I_2	253.8 g I_2	= 40.6 g I_2
	1 mol I_2	

- ii) what percentage yield of iodine was produced.

40.6 g I_2	$\times 100 =$	79.9 %
50.8 g I_2		

2. Zinc and sulphur react to form zinc sulphide according to the equation.



If 25.0 g of zinc and 30.0 g of sulphur are mixed,

- a) Which chemical is the limiting reactant?

25.0 g Zn	1 mol Zn	1 mol ZnS	= 382 mol ZnS	LR
1	65.41 g Zn	1 mol Zn		
30.0 g S	1 mol S	1 mol ZnS	= 934 mol ZnS	XS
1	32.1 g S	1 mol S		

- b) How many grams of ZnS will be formed?

382 mol ZnS	97.51 g ZnS	= 37.2 g ZnS
	1 mol ZnS	

- c) How many grams of the excess reactant will remain after the reaction is over?

$$.934 \text{ mol ZnS} - .382 \text{ mol ZnS} = .552 \text{ mol ZnS}$$

.552 mol ZnS	1 mol S	32.1 g S	= 17.7 g S
	1 mol ZnS	1 mol S	

3.

Limiting Reactant And Percent Yield Answers

Xiang Xie



Limiting Reactant And Percent Yield Answers:

The book delves into Limiting Reactant And Percent Yield Answers. Limiting Reactant And Percent Yield Answers is a vital topic that needs to be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Limiting Reactant And Percent Yield Answers, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Limiting Reactant And Percent Yield Answers
- Chapter 2: Essential Elements of Limiting Reactant And Percent Yield Answers
- Chapter 3: Limiting Reactant And Percent Yield Answers in Everyday Life
- Chapter 4: Limiting Reactant And Percent Yield Answers in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Limiting Reactant And Percent Yield Answers. The first chapter will explore what Limiting Reactant And Percent Yield Answers is, why Limiting Reactant And Percent Yield Answers is vital, and how to effectively learn about Limiting Reactant And Percent Yield Answers.
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4. In chapter 3, the author will examine the practical applications of Limiting Reactant And Percent Yield Answers in daily life. The third chapter will showcase real-world examples of how Limiting Reactant And Percent Yield Answers can be effectively utilized in everyday scenarios.
5. In chapter 4, this book will scrutinize the relevance of Limiting Reactant And Percent Yield Answers in specific contexts. This chapter will explore how Limiting Reactant And Percent Yield Answers is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about Limiting Reactant And Percent Yield Answers. The final chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Limiting Reactant And Percent Yield Answers.

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