

LAB #5 - LIMITING REAGENT**Objective**

- Use stoichiometry to determine the limiting reactant.
- Calculate the theoretical yield.
- Calculate the percent yield of a reaction.

Introduction

In lecture you have learned to read chemical equations and evaluate the mol to mol ratios of reactants and products involved in a chemical reaction. In laboratory experiments it is difficult to measure out chemicals in the exact ratio necessary for the chemical reaction. For time and speed reasons, the reaction mixtures in lab will usually have a limiting and an excess reactant.

Limiting reagent (also called limiting reactant) problems use stoichiometry to determine the theoretical yield for a chemical reaction. The *limiting reactant* will be completely consumed in the reaction and limits the amount of product you can make. The limiting reactant also determines the amount of product you can make (the *theoretical yield*). The reactant that is left over after the reaction is complete is called the *excess reactant*.

Example 1:

Consider the process it takes to make a ham sandwich. You need two slices of bread and one piece of ham to make each sandwich. How many complete sandwiches could you make if you had eighteen slices of bread and six slices of ham? Let's set this up like a chemical equation where the ham and bread are our reactants and the sandwich is our product:



Now we can use stoichiometry to determine the amount of sandwiches we could make if we used all of our reactants.

$$18 \text{ slices of bread} \times \frac{1 \text{ sandwich}}{2 \text{ pieces of bread}} = 9 \text{ sandwiches}$$

$$6 \text{ slices of ham} \times \frac{1 \text{ sandwich}}{1 \text{ slice of ham}} = 6 \text{ sandwiches}$$

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Chemical Energetics, Equilibria and Functional Group Organic Chemistry I - Laboratory Mr. Rohit Manglik, 2024-03-03 EduGorilla Publication is a trusted name in the education sector committed to empowering learners with high quality study materials and resources Specializing in competitive exams and academic support EduGorilla provides comprehensive and well structured content tailored to meet the needs of students across various streams and levels **The Organic Chem Lab Survival Manual** James W. Zubrick, 2016-01-19 Written for the laboratory that accompanies the sophomore junior level courses in Organic Chemistry Zubrick provides students with a valuable guide to the basic techniques of the Organic Chemistry lab The book will help students understand and practice good lab safety It will also help students become familiar with basic instrumentation techniques and apparatus and help them master the latest techniques such as interpretation of infrared spectroscopy The guide is mostly macroscale in its orientation **Chemistry** Carson-Dellosa Publishing, 2015-03-16 Chemistry for grades 9 to 12 is designed to aid in the review and practice of chemistry topics Chemistry covers topics such as metrics and measurements matter atomic structure bonds compounds chemical equations molarity and acids and bases The book includes realistic diagrams and engaging activities to support practice in all areas of chemistry The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help students review and reinforce essential skills in individual science topics The series will be aligned to current science standards *Oswaal CDS Previous Years 12 Solved Question Papers General Knowledge (2018-2023) For 2024 Exam* Oswaal Editorial Board, 2023-10-25 Description of the product 100% updated with Fully Solved Paper of April September 2023 Concept Clarity with detailed explanations of 2018 to 2023 Papers Extensive Practice with 1500 Questions and Two Sample Question Papers Crisp Revision with Mind Maps Expert Tips helps you get expert knowledge master crack CDS in first attempt Exam insights with 5 Year wise 2023 2019 Trend Analysis empowering students to be 100% exam ready *(FREE SAMPLE) NEET 5000+ Chapter-wise SURESHOT Graded Problems in Physics, Chemistry & Biology 2nd Edition* Disha Experts, 2019-11-14 **NEET Chemistry 1500+ MCQs** Disha Experts, 2019-12-24 **Chemistry, Student Study Guide** James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the

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