Workbook answer key

1 A time to remember

Excercise 1

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8

My best friend in school max Michael. He and I were in Mrs. Gilbert's third grade class, and we became friends. We often did crazy things in class, but I don't think Mrs. Gilbert ever really got mad at us. For example, Michael had a pet monkey named Bananas. Sometimes he hid it in Mrs. Gilbert's desk. Later, when she opened the deawer, she always screamed loudly, and the class laughed. After two years, Michael's family moned to another town. We emailed each other for a few years, but then we lost contact. I often wender what he's doing new.

Exercise 2

- SíLvio: No, I wasn't born there, actually. I'm originally from Portugal.
- MARY: That's interesting. So, when <u>did you move</u> to Brazil?
- Sit voo: I moved to Brazil when I was in elementary school.
- MARY: Where did you live?
- Silveo: We lived in Recife. It's a beautiful city in northeast Brazil. Then I went to college.
- MALY: Did you go to school in Recife?
- Sfl.vio: No, I went to school in São Paolo. Mary: And what did you study?
- Silvio: Oh, I studied engineering. But I'm here to go to graduate school.
- MARY: Great! When did you arrive?
- Silvio: I arrived last week. I start school in three days.

 MARY: Well, good luck. And sorry for all the questions!

Expendise 3

Answers will vary.

Expendise 4

200

Rodrigo Santoro is from Petrópolis, near Rio de Janeiro, Brazil.

He is an actor.

100

- False: Rodrigo Santoro studied communication at university.
- 2. True
- 3. True
- False: He was nominated for an MTV Movie Award for use role of King Xerxes.

- False: He plays many different roles, such as a young man in a mental hospital, a king, and a president.
- 6. True

Expendise 5

- 2. My favorite pet was a cat called Felix.
- We used to go to <u>summer camp</u> during our vacations. It was really fun.
- There was a great playground in our neighborhood.
 We used to go there every afternoon.

Exercise 6

- Answers will vary. Possible answers:
- They also used to ride bicycles. Their dog Bruno always used to follow them.
- Kate used to go to the beach every weekend during summer vacation. She hardly ever goes now.
- Peser used to collect comic books. They're now worth a lot of money.
- They used to have a pet rabbit. They don't have any pets now.

Exercise 7

- 2. A: Did you use to collect shells?
 - B: No, we didn't collect shells. We used to build sand castles.
- A: Did you use to swim?
 - B: Yes, we did. We used to swim for hours. Then we played all kinds of sports.
- A: Reality? What kinds of sports did you use to play? or What sports did you use to play?
 - B: Well, we used to play beach volleyball with some other kids.
- 5. At Did you use to lose?
- B: No, we didn't. We used to win!

Exercise 8.

Answers will vary.

Exercise 9:

MARIA: I'm an immigrant here. I was born in Chile and grew up there. I came here in 2005. I wasn't very happy at first. Things were difficult for me. I didn't speak. English, so I went to a community college and studied English there. My English got better and I found this job. What about you?

Exercise 10

- 2. A: Tell me a little about yourself.
- B: What do you want to know?
- 3. A: How old were you when you moved here?
- B: About 16.
- A: Did you learn English here?
 B: No. I studied it in Morocco.
- 5. A: By the way, I'm Lisa.
 - B: Glad to meet you.

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DJ Losen

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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction eas particularly nitration of methyl benzoate go give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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then actual is your actual findings yield actual theoretical x 100 what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 o experiment 3 electrophilic aromatic substitution reactions studocu - Jan 24 2023

the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1 bromo 4 nitrobenzene

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer experiment 11 electrophilic aromatic substitution nitration - Jan 12 2022

on the right the arenium cation has a nitro set with incredibly small 0 6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accommodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4 0ml of nitric acid and 4 0ml of sulfuric acid and will then react with 4 5ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

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lab report 10 nitration of bromobenzene raekwon filmore cm 244 section 40 march 27 2018 introduction for this experiment nitration of bromobenzene was the focus of the lab the benzene is an aromatic compound and when it reacts with wither a mixture of sulfuric acid or nitric acid creates what is known as a nitro group