# Practice Problems: Double Displaceme

1. \_\_\_ Fe(s) + \_\_\_ 
$$H_2SO_4(aq) \rightarrow _{--} Fe_2(SO_4)_3(aq) + ___$$

2. \_\_\_ NaC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>(aq) + \_\_\_ H<sub>2</sub>CO<sub>3</sub>(aq) 
$$\rightarrow$$
 \_\_\_ Na<sub>2</sub>CO<sub>3</sub>(aq

3. \_\_\_ (NH<sub>4</sub>)<sub>3</sub>PO<sub>4</sub>(aq) + \_\_\_ Pb(NO<sub>3</sub>)<sub>4</sub>(aq) 
$$\rightarrow$$
 \_\_\_ Pb<sub>3</sub>(PO<sub>4</sub>

4. \_\_\_ 
$$C_4H_{10}(g) + ___ O_2(g) \rightarrow ___ CO_2(g) + ___ H_2O(g)$$

5. \_\_\_ 
$$H_3PO_4(aq) + ___ HCl(aq) \rightarrow ___ PCl_5(s) + ___ H_2C$$

6. \_\_\_ 
$$Hg(OH)_2(s) + ___ H_3AsO_4(aq) \rightarrow ___ Hg_3(AsO_4)_2(s)$$

7. 
$$HClO_4(aq) + P_4O_{10}(s) \rightarrow H_2PO_4(aq) +$$

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