

Notes Lewis Dot Structures

Representative of only the valence electrons

Valence electrons: electrons in the outermost energy level.

Periodic Table Label:

12345678

Ex: Na



Ex: N



Ex: S



Notes Bonding

Chemical Formula: shows the type and number of atoms in the smallest representative unit of a substance

Subscript	Superscript
Cl_2 Tells the number of elements present: 2 Cls	Cl^{1-} Tells the charge of the ion - means a GAIN of electrons - anion
H_2O H: 2 O: 1	Mg^{2+} + Means a LOSS of electrons - cation
H_3PO_4 H: 3 P: 1 O: 4	CO_3^{2-} A charge on multiple elements means it is a polyatomic ion; the charge is shared over the entire compound

Guided Notes 8 The Covalent Bond Answer Key

Rachel S Tattersall



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