

THE UNIVERSITY OF CHICAGO DEPARTMENT OF CHEMISTRY CHEMISTRY 101: GENERAL CHEMISTRY

LECTURE 1: INTRODUCTION

CHEMISTRY: THE SCIENCE OF MATTER AND ITS TRANSFORMATIONS

Matter is anything that has mass and occupies space. It can be classified into pure substances and mixtures. Pure substances are elements and compounds. Mixtures are combinations of two or more pure substances.

The study of chemistry involves understanding the properties of matter, the changes it undergoes, and the energy associated with these changes.

Chemistry is a central science that bridges physics, biology, and earth science. It provides the foundation for understanding the natural world.

The scientific method is a systematic approach to investigating natural phenomena. It involves making observations, forming hypotheses, conducting experiments, and drawing conclusions.

In chemistry, the scientific method is used to study the properties of matter and the changes it undergoes. It helps us understand the fundamental principles of chemistry.

THE SCIENTIFIC METHOD

The scientific method is a systematic approach to investigating natural phenomena. It involves making observations, forming hypotheses, conducting experiments, and drawing conclusions.

CHEMICAL REACTIONS

A chemical reaction is a process in which one or more substances are converted into one or more new substances.

Chemical reactions are characterized by the breaking and forming of chemical bonds. They involve the transfer of electrons and the release or absorption of energy.

Chemical reactions are classified into different types based on the number of reactants and products, and the nature of the changes.

Understanding chemical reactions is essential for understanding the properties of matter and the changes it undergoes.

Chemical reactions are the basis of many processes in the natural world, including the metabolism of living organisms and the formation of minerals.

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Ying-Ying Zheng



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