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the molar concentration of the solute is sometimes abbreviated by putting square brackets around the chemical formula of
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$$m_{mix} = \frac{m_1 + m_2 + m_3 + \dots}{V_1 + V_2 + V_3 + \dots}$$
 1 5 x 480 1 2 x 520 480 520 720 624 1000 1 344 m q5 with increase in temperature which of these changes a molality b fraction of solute present in water

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