

## Limiting Reagents Worksheet

1. Nitric oxide (NO) reacts with oxygen gas to form nitrogen dioxide (NO<sub>2</sub>), a dark brown gas:



In one experiment 0.866 mol of NO is mixed with 0.503 mol of O<sub>2</sub>.

- Determine the limiting reagent
- Calculate the number of moles of NO<sub>2</sub> produced.

2. The depletion of ozone (O<sub>3</sub>) in the stratosphere has been a matter of great concern among scientists in recent years. It is believed that ozone can react with nitric oxide (NO) that is discharged from high altitude planes. The reaction is



If 7.40 g of O<sub>3</sub> reacts with 0.670 g of NO,

- Which compound will be the limiting reagent?
- How many grams of NO<sub>2</sub> will be produced?
- Calculate the number of moles of the excess reagent remaining at the end of the reaction.

# Limiting Reagent Study Guide

**N Colangelo**



## **Limiting Reagent Study Guide:**

## Unveiling the Power of Verbal Artistry: An Psychological Sojourn through **Limiting Reagent Study Guide**

In a global inundated with displays and the cacophony of immediate communication, the profound power and mental resonance of verbal art often disappear in to obscurity, eclipsed by the regular onslaught of sound and distractions. However, nestled within the lyrical pages of **Limiting Reagent Study Guide**, a fascinating function of literary brilliance that impulses with raw thoughts, lies an unique journey waiting to be embarked upon. Written by a virtuoso wordsmith, this exciting opus books viewers on a psychological odyssey, lightly exposing the latent potential and profound impact stuck within the elaborate internet of language. Within the heart-wrenching expanse of this evocative analysis, we shall embark upon an introspective exploration of the book is key subjects, dissect their captivating writing design, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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