

- 10 In 1859, Thomas Edison invented the nickel-iron battery. The following reaction takes place in the battery.



What is the limiting reactant when 5.00 mol of Fe and 8.00 mol of $\text{Ni}(\text{OH})_2$ react?

How many moles of $\text{Fe}(\text{OH})_2$ will be produced?

How many moles of the other reactant will be in excess?

- 20 One of the first cerium compounds that forms is cerium sesquifluoride (Ce_2CeF_7).



What is the limiting reactant in the reaction of 12.5 mol of cerium fluoride with 50.0 mol of sesquifluoride?

How many moles of Ce_2CeF_7 can be produced?

How many moles of the other reactant will be excess?

- 30 Iron is obtained commercially by the reaction of hematite (Fe_2O_3) with carbon monoxide.



What is the limiting reactant when 25.0 mol of hematite reacts with 30.0 mol of carbon monoxide?

How many grams of Fe can be produced?

Limiting Reactants And Percent Yield 9 3 Answer Key

Thomas Griffiths



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