#### Focus Questions

What types of compounds do you know? Give a few examples.

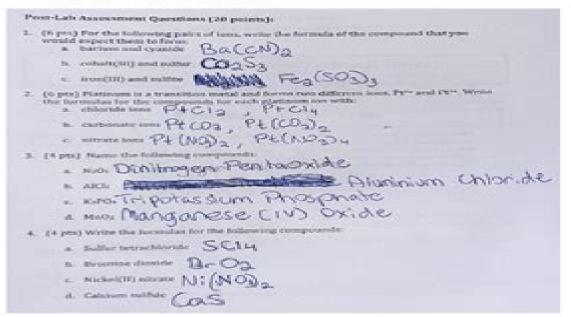
Two examples of compounds are lonic and covalent bonds.

When naming different types of compounds do you need to follow different rules? Explain using examples.

- For naming covalent bonds first write the names of both elements. Then changes the ending of the second one to ide and add the necessary prefixes.
  - SiCl4 → Silicon Tetrachloride
- For ionic compounds, write the metal's name first and then the non-metal. Change non-metals ending to ide. Add roman numerals if necessary.
  - MgBr2 → Magnesium Bromide
  - CuBr → Copper (I) Bromide

3) Does it matter in which order you combine your elements when writing chemical formulas for ionic compounds?

- Yes it does. The cation is typically written first which is the atom which is losing an
  election. This means the position ion is written first and then the negative ion is. The
  polyatomic ion is always written last.
- 4) Which laws govern the formation of compounds?
  - The law of mass action.



# **Hunter Lab Manual Chem 106 Lab**

**S Ashworth** 

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