

Name : \_\_\_\_\_ Date : \_\_\_\_\_



## Gram Formula Mass Worksheet



1. What is the percentage by mass composition of Iron (III) oxide?
2. Calculate the molar mass of  $C_3H_5N_3O_9$  (Nitroglycerin, an explosive)
3. How many atoms are found in 1.55 grams of chlorine gas?
4. When silver sold for \$16.00 per ounce, how many silver atoms could you buy for 10.00 dollars?
5. How many grams of carbon are there in 14.0 g of  $Pb(C_2H_5)_4$  (tetraethyllead, a gasoline additive)?
6. A mixture contains 10.00 g of NaBr and 5.00 g of BaBr<sub>2</sub>. What is the total number of moles of bromide ions in the mixture?
7. Determine the moles of sodium in  $7.22 \times 10^{100}$  kg of  $Na_2S_2O_3$
8. How many atoms of Zn would contain the same number of grams as  $7.54 \times 10^{-6}$  mg of Cu?

# Gram Formula Mass Answer Key

**Yi-Tong Ma**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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