Molarity Calculations

Calculate the molarities of the following solutions:

Maci

2.3 moles of sodium chloride in 0.45 liters of solution.

Ca COa

1.2 moles of calcium carbonate in 1.22 liters of solution.

NINSOU

0.09 moles of sodium sulfate in 12 mL of solution.

LiF

0.75 moles of lithium fluoride in 65 mL of solution.

Ma(CH, COO),5)

0.8 moles of magnesium acetate in 5 liters of solution.

(a(NO2), B)

120 grams of calcium nitrite in 240 mL of solution.

NAOH

 $MW_{G(NV_1)_1} = 132.093/met$ N = 0.91 met7) 98 grams of sodium hydroxide 98 grams of sodium hydroxide in 2.2 liters of solution.

1.2 grams of hydrochloric acid in 25 mL of solution.

45 grams of ammonia in 0.75 L of solution.

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General Chemistry Unit 11 Worksheet 4 Molar Concentration

RS Peters

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