#### UNIT III - THERMOCHEMISTRY

Calculate the H-F bond energy given the following data.

 $H-H(g) + F-F(g) \rightarrow 2 H-F(g)$   $\Delta H_{rxn} = -544 \text{ kJ}$ 

A. 1130 kJ/mol B. 565 kJ/mol C.)283 J/mol D. 300 kJ/mol E. 134 kJ/mol

1.0 g of the rocket fuel hydrazine (N2H4) is burned in a bomb calorimeter. The temperature
of the bomb calorimeter changes from 24.63 °C to 28.16 °C. The heat capacity of the bomb
calorimeter is 837 J/K. Calculate the heat of combustion per mol of hydrazine in the bomb
calorimeter.

A. 95.0 kJ/mol B. 2.96 kJ/mol C. 3.55 x 10<sup>3</sup> B. 1.13 x 10<sup>5</sup> kJ/mol

Calculate the enthalpy change for the following reaction of nitric oxide NO with ozone O3.

 $NO(g) - O_3(g) \rightarrow NO_2(g) + O_2(g)$ 

Given the following data:

NO (g) + O (g)  $\rightarrow$  NO<sub>2</sub> (g)  $\Delta H = -304.1 \text{ kJ}$ 2O<sub>3</sub> (g)  $\rightarrow$  3O<sub>2</sub> (g)  $\Delta H = -284.6 \text{ kJ}$ 2O (g)  $\rightarrow$  O<sub>2</sub> (g)  $\Delta H = -495.0 \text{ kJ}$ A. -1083.7 kJ B. -693.8 kJ C. -198.8 kJ D -93.7 kJ E. -93.7 kJ

4. An endothermic reaction does no work on the surroundings. What values do you predict for the heat and energy change of this reaction?

A. q<0, ΔE>0

D. q>0, ΔE<0

E. q>0, ΔE=0

C. q>0, ΔE>0

 M&M candies contain 209.0. Calories per 1.5 oz serving. The fat content accounts for 38.3% of these Calories. What is the energy of the fat content in kJ?

A. 19.1 kJ B. 874 kJ S 1.91 x 10<sup>-2</sup> kJ D. 3.35 x 10<sup>5</sup> kJ E. 335 kJ

Calculate the heat released when 2.00 g of butane from a cigarette lighter is burned.

2 C<sub>4</sub>H<sub>10</sub> (l) + 13 O<sub>2</sub> (g) - 8 CO<sub>2</sub> (g) + 10 H<sub>2</sub>O (l)

A. -90.7 kJ (B.)-98.3 kJ C. -49.2 kJ D. -108.4 kJ E. -196.6 kJ

7. When Karl Kaveman adds chilled grog to a granite mug, 10.9 kJ of energy is transferred from the mug to the grog. If the mug has a mass of 625 g and an initial temperature of 25 °C, what is the final temperature of the mug? The specific heat of granite is 0.79 J/g·K.

A. 2.9 °C B. 14 °C C. 22 °C D. 25 °C E. 47 °C

# **General Chemistry Multiple Choice Questions And Answers**

Madan R.L.

#### **General Chemistry Multiple Choice Questions And Answers:**

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and teachers Each of the six worldviews results in a unique way of meaning making These six Ways of Knowing or Orders of Consciousness are characterised by increasing complexity of thinking with fourth level thinking or self authorship representing the most common espoused goal of higher education Ample evidence is presented that higher education is not attaining its own espoused goals One explanation may be that many teachers in higher education have not themselves reached the minimum required way of knowing preventing them from constructing a developmental path for their students Van Rossum and Hamer's epistemological model provides clear signposts on the developmental education highway and has proven its worth as an instrument for curriculum design measurement of epistemological development and as a tool for staff development General Chemistry James E. Brady, Gerard E. Humiston, 1986 The Fifth Edition retains the pedagogical strengths that made the previous editions so popular and has been updated reorganized and streamlined Changes include more accessible introductory chapters with greater stress on the logic of the periodic table earlier introduction of redox reactions greater emphasis on the concept of energy a new section on Lewis structures earlier introduction of the ideal gas law and a new development of thermodynamics Each chapter ends with review questions and problems The Vault College Career Bible ,2006 In this annual guide Vault provides overviews of career paths and hiring trends for 2006 in major industries for college graduates Industries covered include accounting banking consulting consumer products and marketing fashion media and entertainment government and politics high tech publishing real estate retail and many more

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