

Chapter 2 - Atoms, Molecules, and Ions

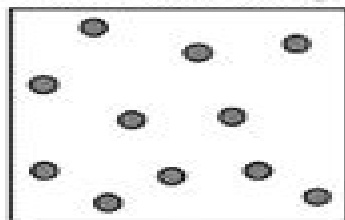
Student: _____

1. Which of the following is/are postulates of Dalton's atomic theory?

1. Atoms combine in fixed ratios of whole numbers.
2. Atoms of each element have different properties.
3. Elements occur as solids, liquids, or gases.

- A. 1 only
B. 2 only
C. 3 only
D. 1 and 2
E. 1, 2, and 3

2. Which of the following statements best describes the particulate representation depicted by the picture?



- A. The figure is a representation of a gas made up of a single element.
B. The figure is a representation of a liquid mixture of two elements.
C. The figure is a representation of a molecular solid.
D. The figure is a representation of a liquid mixture of two compounds.
E. The figure is a representation of a gas of a compound.

3. Which of the following is not a correct name–symbol combination?

- A. cobalt, Co
B. vanadium, V
C. neon, Ne
D. scandium, Sc
E. titanium, Mg

General Chemistry Ebbing 10th Edition Test Bank

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