

## UNIT 1

### 11.1

- Condensed phases- a phase in which the particles are packed in close proximity to one another
- IMFs: solid>liquid>gas
- IMFs are the forces that exist between the molecules, atoms, or ions that make up a particular substance
- Stronger the IMFS, that harder it is to separate the particles.
- Fusion: solid  $\rightarrow$  liquid and Vaporization: liquid  $\rightarrow$  gas both require an input of energy
- Condensation: gas  $\rightarrow$  liquid releases energy
- Enthalpy of vaporization  $\Delta H_{\text{vap}}$  is the energy required to convert one mole of a liquid to vapor at a given temperature
- As IMF  $\uparrow = \Delta H_{\text{vap}} \uparrow$



### 11.2

- Vapor Pressure- the pressure of the vapor of a substance in contact with its liquid or solid phase in a sealed container
  - When liquid is placed in a sealed container, some molecules escape into vapor phase (increasing vapor pressure) and some will recondense into liquid phase. When these become equal, they reach the dynamic equilibrium
  - Equilibrium vapor pressure is the pressure of vapor of a substance at equilibrium in contact with its liquid or solid phase in a sealed container
- $\uparrow$  temperature =  $\uparrow$  VP
- $\uparrow$  IMFs =  $\downarrow$  VP
- Boiling Point of a liquid is when the Vapor pressure of the liquid is equal to the external pressure
  - Normal boiling point is the temperature when the vapor pressure of a liquid is equal to 1 atm (760 mmHg)
- $\uparrow$  IMFs =  $\uparrow$  BP

$$\ln P = \frac{-\Delta H_{\text{vap}}}{RT} + C \quad \ln \frac{P_2}{P_1} = \frac{-\Delta H_{\text{vap}}}{R} \left( \frac{1}{T_2} - \frac{1}{T_1} \right)$$

- $R = 8.314 \times 10^{-3} \text{ kJ/K} \cdot \text{mol}$ ,  $C = \text{constant of integration}$ 
  - $\ln P$  vs.  $1/T$  and the slope is  $-\Delta H_{\text{vap}}/R$

### 11.3

- Cohesive forces: a force of attraction between molecules of a single substance and affect the physical properties of a liquid
- Surface tension is the measure of force required to "break" the surface of a liquid
  - Results from different IMFs experienced by molecules at the surface of a liquid and those in the interior of the liquid

# General Chemistry Semester 2 Study Guide

**Horace Grove Deming, Saul Bryan  
Arenson**



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**Announcements and General Information** Virginia Junior College,1922      Annual Catalogue College of Hawaii,1910

Differentiated Study Guide High School Chemistry Carl David Morneweck,Chester Arthur Buckner,Gale R.

Kirschner,1935      **Directory of Distance Learning Opportunities** Modoc Press, Inc.,2003-02-28 This book provides an overview of current K 12 courses and programs offered in the United States as correspondence study or via such electronic delivery systems as satellite cable or the Internet The Directory includes over 6 000 courses offered by 154 institutions or distance learning consortium members Following an introduction that describes existing practices and delivery methods the Directory offers three indexes Subject Index of Courses Offered by Level Course Level Index Geographic Index All information was supplied by the institutions Entries include current contact information a description of the institution and the courses offered grade level and admission information tuition and fee information enrollment periods delivery information equipment requirements credit and grading information library services and accreditation      **Year-book**

University of Southern California,1906      *Defiance* Defiance College,1914      **Circular of Information** University of Southern California,1904

Exercises in General Chemistry Horace Grove Deming,Saul Bryan Arenson,1924      **General**

**Catalogue** Berea College,1916      AISTSSE 2018 Martina Restuati ,Herbert Sipahutar,Juniastel Rajagukguk,2019-10-04

This book contains the proceedings of the The 5th Annual International Seminar on Trends in Science and Science Education AISTSSE and The 2nd International Conference on Innovation in Education Science and Culture ICIESC where held on 18 October 2018 and 25 September 2018 in same city Medan North Sumatera Both of conferences were organized respectively by Faculty of Mathematics and Natural Sciences and Research Institute Universitas Negeri Medan The papers from these conferences collected in a proceedings book entitled Proceedings of 5th AISTSSE In publishing process AISTSSE and ICIESC were collaboration conference presents six plenary and invited speakers from Australia Japan Thailand and from Indonesia Besides speaker around 162 researchers covering lecturers teachers participants and students have attended in this conference The researchers come from Jakarta Yogyakarta Bandung Palembang Jambi Batam Pekanbaru Padang Aceh Medan and several from Malaysia and Thailand The AISTSSE meeting is expected to yield fruitful result from discussion on various issues dealing with challenges we face in this Industrial Revolution RI 4 0 The purpose of AISTSSE is to bring together professionals academics and students who are interested in the advancement of research and practical applications of innovation in education science and culture The presentation of such conference covering multi disciplines will contribute a lot of inspiring inputs and new knowledge on current trending about Mathematical Sciences Mathematics Education Physical Sciences Physics Education Biological Sciences Biology Education Chemical Sciences Chemistry Education and Computer Sciences Thus this will contribute to the next young generation researches to produce innovative research findings Hopely that the scientific attitude and skills through research will promote Unimed to be a well known university which

persist to be developed and excelled Finally we would like to express greatest thankful to all colleagues in the steering committee for cooperation in administering and arranging the conference Hopefully these seminar and conference will be continued in the coming years with many more insight articles from inspiring research We would also like to thank the invited speakers for their invaluable contribution and for sharing their vision in their talks We hope to meet you again for the next conference of AISTSSE Catalog Washburn University of Topeka,1915 *Subject Guide to Books in Print* ,1997

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## **General Chemistry Semester 2 Study Guide** Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**General Chemistry Semester 2 Study Guide**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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