

# CBSE Class 10 Science Sample Paper Solution

## Section A

1. What are oxidation and reduction reactions? [1]

**Answer 1:** Oxidation is a type of chemical reaction where there is a gain of oxygen or loss of hydrogen while reduction involved loss of oxygen or gain of hydrogen.

2. What happens when an acid reacts with a metal? [1]

**Answer 2:** When an acid reacts with a metal, hydrogen gas is usually released and a salt compound is formed.

## Section B

3. Write the equation for the chemical reaction: Hydrogen + Chlorine → Hydrogen chloride. State whether it is a balanced equation or not? [2]

**Answer 3:**



The 2 in front of HCl indicates that the equation is balanced.

OR

What are covalent Bonds? Name an element which forms such type of bond with itself and other elements?

**Answer:** Bonds that are formed by the sharing of electrons between two atoms are known as covalent bonds. Carbon can form covalent bonds with itself and other elements like hydrogen, nitrogen, oxygen and others.

4. State a reason why rainwater is a good conductor of electricity while distilled water is a bad conductor of electricity? [2]

**Answer 4:** Due to the presence of dissolved salts and ions rainwater is a good conductor of electricity whereas distilled water does not contain ions.

OR

Name one characteristic of metals that allow it to be easily flattened into thin sheets. Give an example of such metals.

**Answer:** Malleability is one physical property of metals that allow it to be shaped into thin sheets. Gold and Silver are some examples of metals having high malleability.

# Fa 1 Sample Paper Class 10 Cbse

**Katrin Zwirgmaier**



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### **Table of Contents Fa 1 Sample Paper Class 10 Cbse**

1. Understanding the eBook Fa 1 Sample Paper Class 10 Cbse
  - The Rise of Digital Reading Fa 1 Sample Paper Class 10 Cbse
  - Advantages of eBooks Over Traditional Books
2. Identifying Fa 1 Sample Paper Class 10 Cbse
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Fa 1 Sample Paper Class 10 Cbse
  - User-Friendly Interface
4. Exploring eBook Recommendations from Fa 1 Sample Paper Class 10 Cbse

- Personalized Recommendations
  - Fa 1 Sample Paper Class 10 Cbse User Reviews and Ratings
  - Fa 1 Sample Paper Class 10 Cbse and Bestseller Lists
5. Accessing Fa 1 Sample Paper Class 10 Cbse Free and Paid eBooks
    - Fa 1 Sample Paper Class 10 Cbse Public Domain eBooks
    - Fa 1 Sample Paper Class 10 Cbse eBook Subscription Services
    - Fa 1 Sample Paper Class 10 Cbse Budget-Friendly Options
  6. Navigating Fa 1 Sample Paper Class 10 Cbse eBook Formats
    - ePub, PDF, MOBI, and More
    - Fa 1 Sample Paper Class 10 Cbse Compatibility with Devices
    - Fa 1 Sample Paper Class 10 Cbse Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Fa 1 Sample Paper Class 10 Cbse
    - Highlighting and Note-Taking Fa 1 Sample Paper Class 10 Cbse
    - Interactive Elements Fa 1 Sample Paper Class 10 Cbse
  8. Staying Engaged with Fa 1 Sample Paper Class 10 Cbse
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Fa 1 Sample Paper Class 10 Cbse
  9. Balancing eBooks and Physical Books Fa 1 Sample Paper Class 10 Cbse
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Fa 1 Sample Paper Class 10 Cbse
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Fa 1 Sample Paper Class 10 Cbse
    - Setting Reading Goals Fa 1 Sample Paper Class 10 Cbse
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Fa 1 Sample Paper Class 10 Cbse



- Fact-Checking eBook Content of Fa 1 Sample Paper Class 10 Cbse
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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