



KEEP IT SIMPLE SCIENCE

Chemistry Module 5

Chemical Equilibrium WORKSHEETS

Worksheet 1

The Basics

Student Name.....

1.
a) When any chemical reaction reaches an equilibrium in a closed system, what becomes true about the concentration of all the chemical species present?

b) Explain why this is so, in the case of a:
i) static equilibrium.

ii) dynamic equilibrium.

2.
What are the thermodynamic reasons why some reactions (eg combustion) are never reversible?

3.
What is meant by "open" or "closed" chemical systems?

4.
Why can a dynamic equilibrium NOT occur in an open system?

5.
Write an ionic equation to represent the dynamic equilibrium reached between the solid and the dissolved ions in each of these species, when a solution reaches saturation.

a) potassium bromide

b) magnesium chloride

c) iron (III) nitrate

d) ammonium sulfate

e) calcium carbonate

Keep It Simple Science Chemistry Answers

John Clarence Hogg



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web the shell thickness is calculated as per the code the effect of along wind load and temperature stresses are only considered for the analysis in this study the chimneys

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the design shear force V and design bending moment M for stack like structure at a distance x from the top of stack shall be calculated as per IS 1893 part 1 see more

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