

Chemistry Chapter 8 Exam

Recall/Identify

Answer each question. Write the best answer below.

- _____ 1. Chemical bonding
- _____ 2. Lewis bonding
- _____ 3. Covalent bonding
- _____ 4. Electronegativity
- _____ 5. Bond strength
- _____ 6. Single bond
- _____ 7. Multiple bonds
- _____ 8. Electron strategy
- _____ 9. Valence shell
- _____ 10. Dipole

- _____ 11. involved by molecules composed of elements that are highly electronegative or polar elements
- _____ 12. change groups of molecules formed atoms
- _____ 13. the energy released when one mole of an ionic crystalline compound is formed from its ions
- _____ 14. related to single bonds
- _____ 15. coordination bonds are part of molecules in which both atoms have lone pairs
- _____ 16. the energy released in bonds is chemical bond and both related to electronegativity
- _____ 17. chemical groups of atoms that are held together by chemical bonds
- _____ 18. related to the ability of elements to form chemical bonds
- _____ 19. chemical bonding that results from the electrical attraction between positive and negative charges
- _____ 20. mutual electrical attraction between protons and electrons, neutrons, and other particles, including the forces between

Multiple-Choice

- _____ 21. The strongest bond is
- (a) ionic bond
 - (b) covalent bond
 - (c) metallic bond
 - (d) hydrogen bond

_____ 22. In many compounds, atoms of electronegative elements form bonds with less electronegative or non-electronegative elements of lower electronegativity.

- (a) 1 (b) 2 (c) 3 (d) 4

_____ 23. Which principle states that atoms tend to form compounds with the most electronegative elements in the periodic table?

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R Bogdan



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