

## Chemical bonding - answers

1. Atoms form molecules by sharing electrons and those are known as covalently bonded molecules. As same, by donating electrons to non-metals from metals the ionic compounds are forms.

a) Group the following compounds – ionic or covalent.

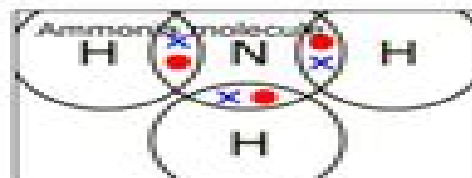
Carbon dioxide      Magnesium      Aluminium oxide      water      sulfur trioxide  
Potassium iodide      methane      ethanol      oxygen      ammonium chloride      ammonia gas  
Diamond      copper nitrate      silicon dioxide      sodium chloride      hydrogen chloride gas

| Covalent compounds | Ionic compounds  |
|--------------------|------------------|
| Carbon dioxide     | Aluminium oxide  |
| Water              | Potassium iodide |
| Sulfur trioxide    | Copper nitrate   |
| Methane            | Sodium chloride  |
| Ethanol            |                  |
| Oxygen             |                  |
| Ammonia gas        |                  |
| Silicon dioxide    |                  |
| Hydrogen gas       |                  |
|                    |                  |

b). Simple molecular compounds can be formed by sharing electrons only. Give 3 examples for simple molecular compounds.

1.....methanol ..... 2.....sulfur dioxide ..... 3.....hydrogen sulfide gas ( any covalent compound)

c. Draw dot cross diagrams to show the bonding in: i) hydrogen chloride molecule      ii) ammonia molecule



d) Explain why the bond in a nitrogen molecule is called a triple bond.

Two nitrogen atoms share three electrons from each atom to form three covalent bonds.

e). How many electrons are need to form a covalent bond?

Two electrons / a pair of electrons

# Holt Mcdougal Activity L1 Chemical Bonding Vocab Activity Answers

**N Colangelo**



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