

Experimental Probability

- **Theoretical probability** is the expected result for an experiment with a very large number of trials.
- **Experimental probability** is the ratio of the number of times an outcome in an event actually occurs to the total number of completed trials.

Complete the table for each probability experiment. Record each probability.

1. Roll two number cubes and get an odd sum, $P(\text{odd sum})$.

Theoretical Probability	Number of Trials	Expected Results	Actual Results	Experimental Probability
$\frac{18}{36} = \frac{1}{2}$	10		odd	

2. Toss 2 coins and get 1 heads and 1 tails, $P(\text{heads, tails})$.

Theoretical Probability	Number of Trials	Expected Results	Actual Results	Experimental Probability
	10		heads, tails	

3. Roll two number cubes and get a sum greater than 7, $P(\text{sum} > 7)$.

Theoretical Probability	Number of Trials	Expected Results	Actual Results	Experimental Probability
	60		sum > 7	

Problem Solving

4. In which experiment were your actual results closest to the expected results?

Show Your Work

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web jan 30 2023 a lewis structure is a very simplified representation of the valence shell electrons in a molecule it is used to show how the electrons are arranged around individual atoms in a molecule electrons are shown as dots or for bonding electrons as a line between the two atoms

lewis dot structures chemistry libretexts - Jan 09 2023

web jan 30 2023 follow these simple steps to draw lewis dot structures draw the atoms on paper and put dots around them to represent valence electrons of the atom be sure to have the correct number of electrons if the species is an ion add or subtract electrons corresponding to the charge of the ion

lewis dot structure practice problems with answers and youtube - Apr 12 2023

web mar 24 2018 practice drawing lewis structures with answers and explanation the video covers the basic lewis structures for a general chemistry class the five steps a