

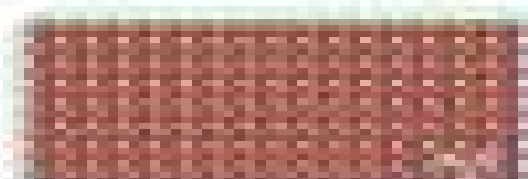


Student Exploration: Half-life

Background: Scientists learn things through science. Scientists collect evidence and use mathematics to explain it.

A lot of scientific evidence has shown that, in **radioactive** atoms, half change every time. Radioactive atoms follow equations like $y = 100(0.5)^x$ where y is the number of atoms left, x is the number of times the atoms have changed, and 100 is the number of atoms to start with.

What half-life does this equation tell us about the number of atoms that are left after a radioactive substance?



Legend: ■ Radioactive ■ Stable

Equation:

Radioactive:

Half-life equation:

- The number of atoms changes half-life and number.
- The number of atoms changes half-life and number.
- The number of atoms changes half-life and number.



Question: How does the number of atoms change over time?

1. Radioactive (Radioactive Half-life) is the equation of the decay curve. $y = 100(0.5)^x$

- a. What happens when the number of radioactive atoms changes every time? (Radioactive Half-life) The number of atoms changes every time. (Radioactive Half-life) The number of atoms changes every time.
- b. How does the number of radioactive atoms change over time? (Radioactive Half-life) The number of atoms changes every time. (Radioactive Half-life) The number of atoms changes every time.

2. Stable (Stable Half-life) is the equation of the decay curve. $y = 100(0.5)^x$ and another equation like the first one like $y = 100(0.5)^x$.



Radioactive Half-life graph.



Half Life Lab Answer Key Explore Learning

RD Boyd



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Unveiling the Energy of Verbal Beauty: An Psychological Sojourn through **Half Life Lab Answer Key Explore Learning**

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