

Musculoskeletal, Excretory and Nervous system

Question 1: Choose the correct option form the choices given below.

[10]

- 1.1 The tough, flexible tissue which attaches bones to bones is called
- A. Tendons
B. Ligaments
C. Cartilage
D. Connective Tissue
- 1.2 The process of all the chemical reactions in the body is
- A. Metabolism
B. Excretion
C. Osmoregulation
D. Respiration
- 1.3 Which of the following is known as an accessory organ of the digestive system?
- A. Pancreas
B. Liver
C. Gall bladder
D. All of the above
- 1.4 A change in the environment that causes a reaction is called a stimulus.
- A. Sensory Organs
B. Receptor
C. Stimulus
D. Effector
- 1.5 The sweat glands in the skin belong to which system?
- A. Respiratory
B. Excretory
C. Nervous
D. Sensory
- 1.6 The function of the axon is to
- A. Carry the impulse away from the cell body.
B. Carry the impulse towards the cell body.
C. Increase the surface area of the cell.
D. Release neurotransmitters.
- 1.7 Which of the following is not a central nervous system stimulant?
- A. Cocaine
B. Ritalin
C. Vicodin
D. Ecstasy
- 1.8 The functional unit of the kidney is the:
- A. Nephron
B. Nephron
C. Hepatic valve
D. Neuron

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Michael Seilmaier



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Table of Contents Grade 9 Midyear Exam Paper Natural Science

1. Understanding the eBook Grade 9 Midyear Exam Paper Natural Science
 - The Rise of Digital Reading Grade 9 Midyear Exam Paper Natural Science
 - Advantages of eBooks Over Traditional Books
2. Identifying Grade 9 Midyear Exam Paper Natural Science
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade 9 Midyear Exam Paper Natural Science
 - User-Friendly Interface
4. Exploring eBook Recommendations from Grade 9 Midyear Exam Paper Natural Science
 - Personalized Recommendations
 - Grade 9 Midyear Exam Paper Natural Science User Reviews and Ratings
 - Grade 9 Midyear Exam Paper Natural Science and Bestseller Lists
5. Accessing Grade 9 Midyear Exam Paper Natural Science Free and Paid eBooks
 - Grade 9 Midyear Exam Paper Natural Science Public Domain eBooks
 - Grade 9 Midyear Exam Paper Natural Science eBook Subscription Services
 - Grade 9 Midyear Exam Paper Natural Science Budget-Friendly Options
6. Navigating Grade 9 Midyear Exam Paper Natural Science eBook Formats
 - ePub, PDF, MOBI, and More
 - Grade 9 Midyear Exam Paper Natural Science Compatibility with Devices
 - Grade 9 Midyear Exam Paper Natural Science Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Grade 9 Midyear Exam Paper Natural Science
 - Highlighting and Note-Taking Grade 9 Midyear Exam Paper Natural Science
 - Interactive Elements Grade 9 Midyear Exam Paper Natural Science
8. Staying Engaged with Grade 9 Midyear Exam Paper Natural Science

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Grade 9 Midyear Exam Paper Natural Science
- 9. Balancing eBooks and Physical Books Grade 9 Midyear Exam Paper Natural Science
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Grade 9 Midyear Exam Paper Natural Science
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Grade 9 Midyear Exam Paper Natural Science
 - Setting Reading Goals Grade 9 Midyear Exam Paper Natural Science
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Grade 9 Midyear Exam Paper Natural Science
 - Fact-Checking eBook Content of Grade 9 Midyear Exam Paper Natural Science
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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 $\text{Pb} + \text{HCl} \rightarrow \text{PbCl}_2 + \text{H}_2$
 $\text{AlBr}_3 + \text{K}_2\text{SO}_4 \rightarrow 6\text{KBr} + \text{Al}_2\text{SO}_4$
 $\text{CH}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
 $\text{C}_3\text{H}_8 + 5\text{O}_2 \rightarrow 3\text{CO}_2 + 4\text{H}_2\text{O}$
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 $\text{Mg} + 2\text{Li}_2\text{CO}_3 \rightarrow \text{MgCO}_3 + 2\text{Li}_2\text{O}$
 $\text{LiF} + \text{P}_4\text{O}_{10} \rightarrow 2\text{Li}_2\text{O} + 3\text{P}_2\text{O}_5$
 $\text{RbNO}_3 \rightarrow \text{RbNO}_2 + \text{O}_2$
 $\text{Be}_3\text{N}_2 + 6\text{H}_2\text{O} \rightarrow 3\text{Be}(\text{OH})_2 + 2\text{NH}_3$

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 $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$
 $\text{HNO}_3 + \text{H}_2\text{O} \rightarrow \text{H}_2\text{O} + \text{O}_2$
 $\text{H}_2\text{O} + \text{O}_2 \rightarrow \text{H}_2\text{O}$
 $\text{RbH}_2\text{O} + \text{RbOH} \rightarrow \text{H}_2\text{O} + \text{Cl}_2$
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 $\text{H}_2 + \text{NH}_3 \rightarrow \text{N}_2 + \text{H}_2\text{O}$
 $\text{Ca} + \text{N}_2 \rightarrow \text{CaF}_2 + \text{CH}_4$
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 $2\text{Fe} + 3\text{H}_2\text{SO}_4 \rightarrow \text{Fe}_2(\text{SO}_4)_3 + 3\text{H}_2$
 $2\text{C}_4\text{H}_{10} + 13\text{O}_2 \rightarrow 8\text{CO}_2 + 10\text{H}_2\text{O}$

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 $3\text{Fe} + 2\text{O}_3 \rightarrow 3\text{FeO}$
 $2\text{Fe} + 3\text{CO} \rightarrow 2\text{Fe}_3\text{O}_4 + 3\text{CO}_2$
 $3\text{BaO} + \text{Ba} \rightarrow 4\text{Ba} + 2\text{H}_2\text{O}$
 $40\text{Mg} + \text{NH}_4\text{PO}_4 \rightarrow 4\text{Mg}_2\text{P} + 2\text{O}_7$
 $2\text{NH}_3 + \text{H}_2\text{O} \rightarrow 4\text{H} + 2\text{Al}$
 $3\text{H}_2\text{SO}_4 + 3\text{H}_2 \rightarrow 4\text{Al}_2\text{SO}_4 + 3\text{H}_2$
 $4\text{P} + 10\text{HCl} \rightarrow 3\text{HPO}_3 + 4\text{H}_2\text{O} + 3\text{O}_2$
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