

STUDY GUIDE: Understanding Properties of Matter

1. Use examples to describe the difference between physical and chemical properties of matter. (See handout)
2. List 9 physical properties of matter with an example of each. Density, Ductility, Electrical Conductivity, Magnetism, Malleability, Solubility, Specific Heat, State, Thermal Conductivity (Examples will vary but should be similar to the examples found in the Properties of Matter Booklet.)
3. List 2 chemical properties of matter with an example of each. Flammability, Reactivity (Examples will vary but should be similar to the examples found in the Properties of Matter Booklet.)
4. Which property of matter is observed when water changes from a solid to a liquid? State
5. Adding food coloring to the solution is an example of solubility.
6. If bubbles start to rise from a solution in the scientist's test tube shown, it would be an indication of reactivity.
7. What "point" is observed when water changes from a solid to a liquid? Melting point
8. Which property(ies) would you use to describe most metals? Ductility or Malleability or Electrical Conductivity
9. Which of the properties of metals means that it can be drawn into wires? Ductility
10. Which property allows metals to be bent rather than broken? Malleability
11. Besides being valuable, gold is special because it can be hammered into a very thin foil. This property is called malleability.

Use the table below to interpret questions 12 – 15.

SUBSTANCE	DENSITY (g/cm ³)	SUBSTANCE	DENSITY (g/cm ³)
Aluminum (solid)	2.7	Ice (solid)	0.93
Pyrite (solid)	5.02	Water (liquid)	1.00
Mercury (liquid)	13.55	Zinc (solid)	7.13
Silver (solid)	10.50	Wood (oak)	0.85

12. What substance has a density more than 13 times greater than water? Mercury
13. What is the density of ice? .93g/cm³
14. Which metal is a liquid at room temperature? Mercury
15. What two variables determine the density of a substance? Mass/volume
16. The electromagnets found inside of doorbells are an example of which properties of matter? Magnetism; Electrical conductivity; Ductility (coiled wire around metal)
17. Chemical weathering, such as acid rain causing a statue to fizz and deteriorate is an example of which property of matter? Reactivity
18. Which property of matter is exemplified by pressing metals into coins? Malleability
19. Which property of matter is observed when water changes from a liquid to a solid? State
20. What "point" is observed when water changes from a liquid to a solid? Freezing point
21. Which property of matter is observed when water changes from a liquid to a gas? State
22. What "point" is observed when water changes from a liquid to a gas? Boiling point

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Virginia M. Tucker, Marc Lampson

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the authors' names.

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Table of Contents **Matter Of States Viewing Guide Key**

1. Understanding the eBook **Matter Of States Viewing Guide Key**
 - The Rise of Digital Reading **Matter Of States Viewing Guide Key**
 - Advantages of eBooks Over Traditional Books
2. Identifying **Matter Of States Viewing Guide Key**
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an **Matter Of States Viewing Guide Key**
 - User-Friendly Interface
4. Exploring eBook Recommendations from **Matter Of States Viewing Guide Key**
 - Personalized Recommendations
 - **Matter Of States Viewing Guide Key** User Reviews and Ratings
 - **Matter Of States Viewing Guide Key** and Bestseller Lists

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 - Matter Of States Viewing Guide Key eBook Subscription Services
 - Matter Of States Viewing Guide Key Budget-Friendly Options
6. Navigating Matter Of States Viewing Guide Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Matter Of States Viewing Guide Key Compatibility with Devices
 - Matter Of States Viewing Guide Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Matter Of States Viewing Guide Key
 - Highlighting and Note-Taking Matter Of States Viewing Guide Key
 - Interactive Elements Matter Of States Viewing Guide Key
8. Staying Engaged with Matter Of States Viewing Guide Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Matter Of States Viewing Guide Key
9. Balancing eBooks and Physical Books Matter Of States Viewing Guide Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Matter Of States Viewing Guide Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Matter Of States Viewing Guide Key
 - Setting Reading Goals Matter Of States Viewing Guide Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Matter Of States Viewing Guide Key
 - Fact-Checking eBook Content of Matter Of States Viewing Guide Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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