

Half-Life

#4

If 100.0 g of carbon-14 decays until only 25.0 g of carbon is left after 11,460 y, what is the half-life of carbon-14?

1. List the given and unknown values.

Given: initial mass of sample = 100.0 g
 final mass of sample = 25.0 g
 total time of decay = 11 460 y

Unknown: number of half-lives = ? half-lives
 half-life = ? y

2. Write down the equation relating half-life, the number of half-lives, and the decay time, and rearrange it to solve for half-life.

$$\text{total time of decay} = \text{number of half-lives} \times \frac{\text{number of years}}{\text{half-life}}$$

$$\frac{\text{number of years}}{\text{half-life}} = \frac{\text{total time of decay}}{\text{number of half-lives}}$$

3. Calculate how many half-lives have passed during the decay of the 100.0 g sample.

$$\text{fraction of sample remaining} = \frac{\text{final mass of sample}}{\text{initial mass of sample}} = \frac{25.0 \text{ g}}{100.0 \text{ g}} = \frac{1}{4}$$

$$\text{after one half-life} = \frac{1}{2}; \text{ after two half-lives} = \frac{1}{2} \times \frac{1}{2} = \frac{1}{4} \text{ of sample}$$

Two half-lives have passed.

4. Solve for the half-life.

$$\frac{\text{number of years}}{\text{half-life}} = \frac{11\,460 \text{ y}}{2 \text{ half-lives}} = \frac{5730 \text{ y}}{\text{half-life}}$$

$$\text{half-life of carbon-14} = 5730 \text{ y}$$

Practice

- What is the half-life of a 100.0 g sample of nitrogen-16 that decays to 12.5 g of nitrogen-16 in 21.6 s?
- All isotopes of technetium are radioactive, but they have widely varying half-lives. If an 800.0 g sample of technetium-99 decays to 100.0 g of technetium-99 in 639 000 y, what is its half-life?
- A 208 g sample of sodium-24 decays to 13.0 g of sodium-24 within 60.0 h. What is the half-life of this radioactive isotope?

Math Skills Half Life Answers

Michal Rosen-Zvi



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In a world inundated with screens and the cacophony of fast interaction, the profound power and emotional resonance of verbal beauty usually disappear in to obscurity, eclipsed by the constant onslaught of sound and distractions. However, nestled within the musical pages of **Math Skills Half Life Answers**, a captivating perform of fictional elegance that impulses with organic emotions, lies an unique journey waiting to be embarked upon. Composed with a virtuoso wordsmith, this magical opus guides readers on an emotional odyssey, delicately revealing the latent possible and profound affect embedded within the complex web of language. Within the heart-wrenching expanse with this evocative examination, we will embark upon an introspective exploration of the book is central themes, dissect their interesting publishing style, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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