

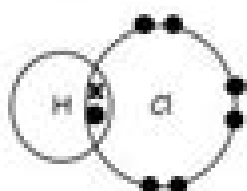
C3 - STRUCTURE AND BONDING 3.5 Covalent bond

Aiming for Grade 4

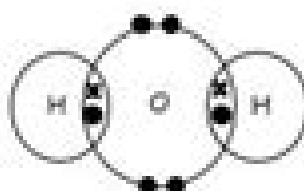
Ionic bonds form between **metals** and **non-metals**. Covalent bonds form between **non-metals**. Covalent bonds form when atoms **share** pairs of **electrons**. These bonds are very **strong**.

Share, Metals, strong, non-metals, electrons
You can use the words more than once.

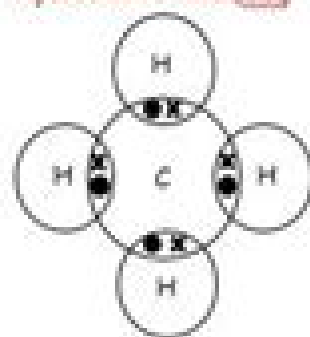
Name these small molecules and state their formulae.



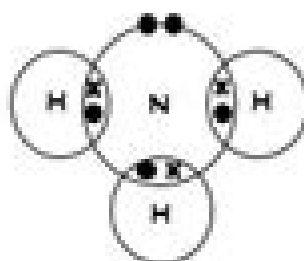
Hydrochloric Acid (HCl)



Water (H_2O)



Methane (CH_4)



Ammonia (NH_3)

Assessment

Aiming for Grade 6

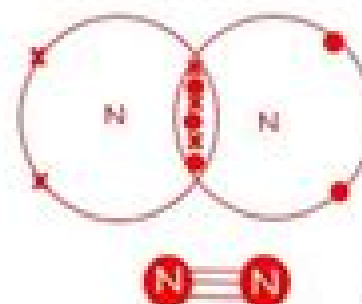
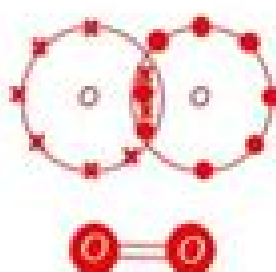
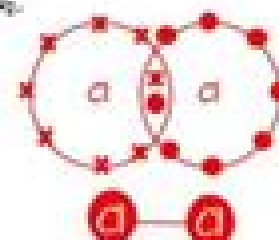
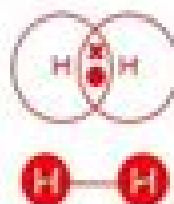
Describe how Hydrogen and Chlorine form a covalent bond to form Hydrochloric acid.

To gain full outer shells Hydrogen needs to gain an electron and Chlorine needs to lose one. Hydrogen shares its outer shell electron with Chlorine and Chlorine shares one of its electrons.

What is a double covalent bond?

Two shared pairs of electrons

Draw dot and cross diagrams and ball and stick diagrams for H_2 , Cl_2 , O_2 , and N_2 .



Assessment

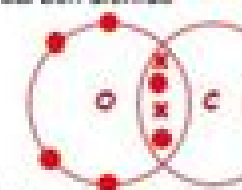
Aiming

Draw ball and stick or the following covalent

Three hydrogen atoms
one phosphorus (phos)



Carbon dioxide



Ethane (2 carbon with hydrogen)



How do double and triple properties?

A double or triple bond the bond making the me

Assessment
Grade 6

1. I can draw dot and

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Yijin Wang



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In a global consumed by monitors and the ceaseless chatter of immediate interaction, the melodic splendor and emotional symphony developed by the written term frequently diminish into the backdrop, eclipsed by the constant noise and interruptions that permeate our lives. But, located within the pages of **June 2013 Aqa Chemistry C3 Mark Scheme** an enchanting literary treasure filled with natural thoughts, lies an immersive symphony waiting to be embraced. Crafted by an outstanding musician of language, this fascinating masterpiece conducts visitors on a psychological journey, skillfully unraveling the hidden tunes and profound affect resonating within each carefully constructed phrase. Within the depths with this moving review, we will investigate the book is central harmonies, analyze their enthralling writing model, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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