

Half-life problems

1. Fluorine-21 has a half-life of approximately 5 seconds. What percentage of the original nuclei would remain after 20 seconds?
2. Iodine-131 has a half-life of 8 days. How many grams of a 48 gram sample would remain at the end of 32 days?
3. The half-life of chromium-51 is 28 days. If the sample contained 5.10 grams, how much chromium was present 168 days ago?
4. If 20.0 g of a radioactive isotope are present at 1:00 PM and 5.0 g remain at 2:00 PM, what is the half-life of the isotope?
5. The half-life of Uranium-238 is 4.6 billion years and the age of earth is 4.6 billion years. What percentage of Uranium-238 that was present when Earth was formed still remains?
6. Chromium-48 decays. After 3 half-lives, what percentage of the original nuclei would remain?
7. The half-life of iodine-125 is 60 days. About how much of a 144 gram sample will remain after one year?
8. Titanium-51 decays with a half-life of 6 minutes. How many minutes will it take for 480 grams of titanium-51 to decay down to just 15 grams?
9. A medical institution requests 1 g of bismuth-214, which has a half-life of 20 min. How many grams of bismuth-214 must be prepared if the shipping time is 2 h?
10. The half-life of radium-226 is 1602 years. If you have 500 grams of radium today how many grams would have been present 3204 years ago?

Answers:

- | | | | | | |
|-----------|--------------|------------|------------|--------|----------|
| 1. 6.25% | 2. 3.0 grams | 3. 326.4 g | 4. 30 min | 5. 50% | 6. 12.5% |
| 7. 2.25 g | 8. 30 min. | 9. 64 g | 10. 2000 g | | |

Half Life Calculations Worksheet Answer Key Pg 100

Nat Reed



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Table of Contents Half Life Calculations Worksheet Answer Key Pg 100

1. Understanding the eBook Half Life Calculations Worksheet Answer Key Pg 100
 - The Rise of Digital Reading Half Life Calculations Worksheet Answer Key Pg 100
 - Advantages of eBooks Over Traditional Books
2. Identifying Half Life Calculations Worksheet Answer Key Pg 100
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Half Life Calculations Worksheet Answer Key Pg 100
 - User-Friendly Interface
4. Exploring eBook Recommendations from Half Life Calculations Worksheet Answer Key Pg 100
 - Personalized Recommendations
 - Half Life Calculations Worksheet Answer Key Pg 100 User Reviews and Ratings
 - Half Life Calculations Worksheet Answer Key Pg 100 and Bestseller Lists
5. Accessing Half Life Calculations Worksheet Answer Key Pg 100 Free and Paid eBooks

- Half Life Calculations Worksheet Answer Key Pg 100 Public Domain eBooks
 - Half Life Calculations Worksheet Answer Key Pg 100 eBook Subscription Services
 - Half Life Calculations Worksheet Answer Key Pg 100 Budget-Friendly Options
6. Navigating Half Life Calculations Worksheet Answer Key Pg 100 eBook Formats
- ePub, PDF, MOBI, and More
 - Half Life Calculations Worksheet Answer Key Pg 100 Compatibility with Devices
 - Half Life Calculations Worksheet Answer Key Pg 100 Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Half Life Calculations Worksheet Answer Key Pg 100
 - Highlighting and Note-Taking Half Life Calculations Worksheet Answer Key Pg 100
 - Interactive Elements Half Life Calculations Worksheet Answer Key Pg 100
8. Staying Engaged with Half Life Calculations Worksheet Answer Key Pg 100
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Half Life Calculations Worksheet Answer Key Pg 100
9. Balancing eBooks and Physical Books Half Life Calculations Worksheet Answer Key Pg 100
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Half Life Calculations Worksheet Answer Key Pg 100
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Half Life Calculations Worksheet Answer Key Pg 100
- Setting Reading Goals Half Life Calculations Worksheet Answer Key Pg 100
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Half Life Calculations Worksheet Answer Key Pg 100
- Fact-Checking eBook Content of Half Life Calculations Worksheet Answer Key Pg 100
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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