

1. Define half-life. *(1 mark)*
2. How is the half-life of a radioisotope similar to a sporting tournament in which the losing team is eliminated? *(2 marks)*
3. The half-life of radium-226 is 1600 years. If a sample of radium-226 has an original activity of 200 Bq, what will its activity be after 4800 years? *(2 marks)*
4. Sodium-24 has a half-life of 15 hours. If a sample of sodium-24 has an original activity of 500 Bq, what will its activity be after 60 hours? *(2 marks)*
5. After 42 days the activity of a sample of phosphorus-32 has decreased from 400 Bq to 50 Bq. What is the half-life of phosphorus-32? *(2 marks)*
6. The half-life of radon-222 is 3.8 days. What was the original activity if it has an activity of 10 Bq after 7.6 days? *(2 marks)*
7. The half-life of thorium-227 is 19 days. How many days are required for 75% of a sample to decay? *(2 marks)*
8. The half-life of protactinium-234 is 6.75 hours. What percentage of a sample will remain after 27 hours? *(2 marks)*
9. A rock once contained 1.0 mg of uranium-238, but now contains only 0.25 mg. Given that the half-life for uranium-238 is 4.5×10^9 (4.5 billion) years, how old is the rock? *(2 marks)*
10. The half-life of tritium (hydrogen-3) is 12.3 years. If 48.0 mg of tritium is released from a nuclear power plant during the course of a mishap, what mass of the sample will remain after 49.2 years? *(2 marks)*

Total marks: 19

Half Life Calculations Instructional Fair

Trevor Pinch, Karin Bijsterveld



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Chemistry Homework Frank Schaffer Publications, Joan DiStasio, 1996-03 Includes the periodic table writing formulas balancing equations stoichiometry problems and more Smith's Inductive Arithmetic, and Federal Calculator Roswell Chamberlain Smith, 1854 **Foundations of College Chemistry** Morris Hein, Susan Arena, Cary Willard, 2023 Foundations of College Chemistry 16th edition presents chemistry as a modern vital subject and is designed to make introductory chemistry accessible to all beginning students It is intended for students who have never taken a chemistry course or those who had a significant interruption in their studies but plan to continue with the general chemistry sequence The central focus is to make chemistry interesting and understandable and teach students the problem solving skills they will need This International Adaptation offers new and updated content with improved presentation of all course material It builds on the strengths of previous editions including clear explanations and step by step problem solving The material emphasizes real world applications of chemistry as the authors develop the principles that form the foundation for the further study of chemistry There is new and expanded coverage of polarizing power and polarizability Fajans rules collision number and mean free path abnormal molecular masses and van t Hoff factor and applications of radioactivity **Physics Teacher Education** Joan Borg Marks, Pauline Galea, Suzanne Gatt, David Sands, 2022-09-15 This book presents the most up to date research contributions focusing on progress in the field of physics education It provides researches and results that are based on the most relevant matters in physics teacher education and how these matters can be improved for the satisfaction of both teachers and learners The work is the by product of the collaboration between GIREP the International Research Group on Physics Teaching and the University of Malta The contributing authors present close examinations of the following topics ICT and multimedia in teacher education experiments and laboratory work in teacher education the role of quantum mechanics in teaching and learning physics formal non formal and informal aspects of physics education at the primary level strategies for pre service physics teacher education at all levels and in service teacher professional learning strategies The editors hope that many different stakeholders within scientific academia will find something of value in this compilation of the current most advanced ideas in physics education Precalculus Robert F. Blitzer, Wayne J. Bentley, 2003 Gets Them Engaged Keeps Them Engaged Blitzer s philosophy present the full scope of mathematics while always 1 engaging the student by opening their minds to learning 2 keeping the student engaged on every page 3 explaining ideas directly simply and clearly so they don t get lost when studying and reviewing The Mirror Of Literature, Amusement, And Instruction; Volume 5 Reuben Percy, John Timbs, 1838 This work has been selected by scholars as being culturally important and is part of the knowledge base of civilization as we know it This work was reproduced from the original artifact and remains as true to the original work as possible Therefore you will see the original copyright references library stamps as most of these works have been housed in our most important libraries around the world and other notations in the work This work is in the public

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