Radioactive Isotopes Half-life MCQ Worksheet

1. When a radioactive isotope decays over time, its half-life

a. Decreases

b. Increases

c. Stays the same

2. As the temperature of a radioactive isotope decreases, its half-life

a. Decreases

b. Increases

c. Stays the same

3. A sample of radioactive isotope that initially had a weight of 80 milligrams. It decays to 5 milligrams after 32 days. What is its half-life?

a. 8 days

b. 12 days

c. 4 days

4. The half-life of iodine-131 is 8.07 days. How much of a sample of iodine-131 is left after 24.21 days?

a. ½

b. 3/5

C. 1/8

5. What is the ratio of uranium-238 to lead-206 in a mineral used to determine?

a. Age

b. Density

c. Solubility

6. What is the isotope carbon-14 used for?

Controlling fission reactions

b. Determining the age of the sample

c. controlling speeds of neutrons

7. An isotope has a half-life of 2.5 minutes. What percentage of the isotope is left after 10 minutes?

a. 50 %

b. 3.125%

c. 6.25 %

How many milligrams of tritium will remain after 49.2 years if the starting amount is 32 mg? The half-life of tritium is 12.3 years.

a. Age

b. Density

c. Solubility

Half Life Of Radioactive Isotopes Worksheet Answers

LL Leslie

Half Life Of Radioactive Isotopes Worksheet Answers:

Reviewing Half Life Of Radioactive Isotopes Worksheet Answers: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Half Life Of Radioactive Isotopes Worksheet Answers**," an enthralling opus penned by a highly acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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