

5. Analyze: What pattern, if any, do you see in your data?

I hardly notice any differences save the general decline.

6. Revise and repeat: Use your data from #4 above to fill in the first line of the data table below. Then repeat the experiment four more times. Calculate the average number of radioactive atoms for each time.

Trial	0 s	10 s	20 s	30 s	40 s	50 s
1	128	55	28	11	6	3
2	128	71	33	17	11	7
3	128	58	27	16	4	3
4	128	64	32	17	9	5
5	128	60	31	12	5	2
Averages:	128	62	30	15	7	4

7. Analyze: A **half-life** is defined as the amount of time it takes for half of the radioactive particles to decay. For the simulated substance, every 10 seconds represents one half-life.

How does your data demonstrate the definition of a half-life?

On average, it is half of the previous one.

8. Revise and repeat: Click **Reset**. Real radioactive samples will contain billions of radioactive atoms. To model the decay of a large sample, change from **Random decay** to **Theoretical decay** on the SIMULATION pane. Click **Play** and record the numbers of radioactive atoms:

0 s: 128 10 s: 64 20 s: 32 30 s: 16 40 s: 8 50 s: 4

How does this data demonstrate the meaning of half-life?

Every life is exactly half of the previous one.

Half Life Gizmo Answer Key

Jin-Ying Zhang

A red circular graphic with a gradient, appearing as a stylized arrow or a light effect, positioned to the right of the author's name.

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