#### **Gravitational Potential Energy Practice Problems**

A man with a mass of 80 kg is standing on the edge of a table that is 1.5
meters above the ground. What is the man's potential energy?

GPE=mgh = 80 kg x 9.8 m/s 2 x 1.5 m = 11.1765

A high jumper clears a vertical distance of 2.45 m, and at the top of his trajectory his potential energy was 1590 J. What is the jumper's mass?

> [m = 66.2 kg] m = 66.2 kg

 A city is situated 4080 m above sea level. If a car with a mass of 905 kg is driven from a location that is 1860 m above sea level to the city, what is the increase in potential energy?

905 kg x 9.8 m/s 2 x 9080 m = 36,185,520 J 905 kg x 9.8 m/s 2 x 1860 m = 16,496,3405 36,185,520 J - 16,496,340 J = 19.689,180 J

4. A 3.0 kg bowling ball is pushed down a 15 m high ramp. What is the potential energy of the bowling ball at the top of the ramp, in the middle of the ramp, and at the bottom of the ramp?

Top= 3 kg x 9.8 m/s 2 x 15 m = 441 5 niddle= 3 kg x 9.8 m/s 2 x 7.5 m = 220.55 Bottom= 3 kg x 9.8 m/s 2 x 0 m = 05

What relationship exists between gravitational potential energy and the height above the ground? Why?

As Deign+ 1. GPET

Challenge

6. A 2.0 kg rock is nudges off the edge of a 20 m ledge. At some point during the fall, the rock has lost 250 J of potential energy. At what height does this occur?

> GPE=mgn = 2kg x 9.8mls2x20m=3925 3925-2505 = 1425 1425 = mgn = 2 kg x 9.8mls2x H [H = 7.24m]

# **Energy Practice Problems With Key**

**CL Gary** 

## **Energy Practice Problems With Key:**

Common Entrance 13+ Science Exam Practice Questions and Answers Ron Pickering, 2022-05-13 Exam board ISEB Level 13 CE and KS3 Subject Science First exams November 2022 Improve exam technique and build confidence to ensure success in the CE 13 Science exams with these practice questions in the style of the ISEB exams Practice for all sections of the exam Biology Chemistry and Physics for Foundation and Level 2 papers Guidance on the exam how to approach different types of questions and understand what is required from all possible questions Identify gaps in knowledge answers are included to help pinpoint areas for improvement Pair with Common Entrance 13 Science Revision Guide for comprehensive exam Survival Guide to General Chemistry Patrick E. McMahon, Rosemary preparation ISBN 9781398340923 McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics. The focus is not to recap or review the theoretical concepts well described in the available texts Instead the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester Each is written in a specific and detailed step by step process for problem solving whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid base reactions which often persists through working with acid base equilibrium Many chapters provide alternative viewpoints as an aid to understanding This book addresses a very real need for a large number of incoming freshman in STEM fields

Thermodynamics Problem Solving in Physical Chemistry Kathleen E. Murphy, 2020-03-23 Thermodynamics Problem Solving in Physical Chemistry Study Guide and Map is an innovative and unique workbook that guides physical chemistry students through the decision making process to assess a problem situation create appropriate solutions and gain confidence through practice solving physical chemistry problems The workbook includes six major sections with 20 30 solved problems in each section that span from easy single objective questions to difficult multistep analysis problems Each section of the workbook contains key points that highlight major features of the topic to remind students of what they need to apply to solve problems in the topic area Key Features Provides instructor access to a visual map depicting how all equations used in thermodynamics are connected and how they are derived from the three major energy laws Acts as a guide in deriving the correct solution to a problem Illustrates the questions students should ask themselves about the critical features of the concepts to solve problems in physical chemistry Can be used as a stand alone product for review of Thermodynamics

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