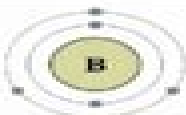


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Review – Bohr & Lewis Dot Diagrams

1. Complete the chart below: Remember that the Bohr diagram shows the number of electrons in all shells, while Lewis dot structures only show the element symbol surrounded by the correct number of valence electrons. The element family & number of valence electrons are found by looking on the periodic table.

Element	Bohr Diagram	Name of Family	# of Valence Electrons	Lewis Dot Structure
Boron		Metalloids or Non-metals	3	$\cdot \dot{\text{B}} \cdot$
Carbon	1 st shell 2 electrons 2 nd shell 4 electrons	Non-metals	4	$\cdot \dot{\text{C}} \cdot$
Oxygen	1 st shell 2 electrons 2 nd shell 6 electrons	Non-metals	6	$\cdot \ddot{\text{O}} \cdot$
Fluorine	1 st shell 2 electrons 2 nd shell 7 electrons	Halogens	7	$:\ddot{\text{F}}:$
Neon	1 st shell 2 electrons 2 nd shell 8 electrons	Noble Gases	8	$:\ddot{\text{Ne}}:$
Sodium	1 st shell 2 electrons 2 nd shell 8 electrons 3 rd shell 1 electron	Alkali Metals	1	$\text{Na} \cdot$
Aluminum	1 st shell 2 electrons 2 nd shell 8 electrons 3 rd shell 3 electrons	Poor Metals	3	$\cdot \dot{\text{Al}} \cdot$

Lewis Structure Answer Key

Guillaume Favre

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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