

# MATHEMATICS

## FIRST TERM JSS 3

WEEK	TOPIC / CONTENT	ACTIVITIES
1	<b>BINARY NUMBERS</b> i. Addition and subtraction of binary numbers ii. Multiplication and division of binary numbers iii. Solving Quantitative Reasoning problems on binary numbers.	<b>Students:</b> i. Add binary numbers ii. Subtract binary numbers iii. Multiply binary numbers iv. Divide given binary numbers v. solve Quantitative Reasoning problems  <b>Instructional Resources:</b> Flash cards
2	<b>BINARY NUMBERS CONT.</b> i. Conversion from base ten to binary ii. Conversion from binary to other bases. iii. Applying binary numbers as two way classification system using punch card. iv. Using computer to do simple mathematical calculations.	<b>Students</b> a. change numbers from base 10 to base 2. b. change base 2 numbers to base 10 or any required base.  <b>Instructional Resources:</b> a. Flash cards showing a typical conversion from base 10 to base 2. b. Punch cards.
3	<b>DIRECT AND INVERSE PROPORTION</b> i. Direct proportion ii. Indirect proportion iii. Apply direct and inverse proportions to practical problems.	<b>STUDENTS:</b> i. Solve problems on direct and inverse proportion including practical problems.  <b>Instructional Resources:</b> Direct and inverse proportions chart
4	<b>RATIONAL AND NON-RATIONAL NUMBERS</b> i. Identifying rational and non-rational numbers. ii. Determining the approximate value of some non-rational numbers. iii. Determining the approximate value of $\pi$ . iv. Finding reciprocals.	<b>Students:</b> a. Identify rational numbers amongst a set of given numbers. b. Determine practically the approximate value of some non-rational numbers.  <b>Instructional Resources:</b> Chart showing some non-rational numbers.
5	<b>PLANE FIGURES</b> Problems in measurement involving: i. Area of triangles ii. Area of parallelograms iii. Area of trapezium iv. Area of circles and sectors v. Word problems involving area.	<b>Students:</b> a. Derive formula for area of triangles. b. Use formula to find the area of parallelograms, trapezia and circles. c. Solve Quantitative problems on areas.  <b>Instructional Resources:</b> Models of indicated shapes (Triangles, Circles etc.)
6	<b>FACTORIZATION OR ALGEBRAIC EXPRESSION</b> Factorization of expressions of the form: i. $ax+ay$ ii. $3m+pq+3p+mq$ iii. $a^2 - b^2$ iv. $a^2 - 2ab + b^2$ v. Word problems involving factorization.	<b>Students:</b> a. Factorize simple algebraic expressions b. Use the quadratic equation box to factorize quadratic expression.  <b>Instructional Resources:</b> a. Quadratic equation box b. Flash cards of problems
7	<b>EQUATIONS INVOLVING FRACTIONS</b>	<b>Students:</b>

# Jss 1 Third Term Mathematics Scheme Of Work

**Sibashis Nanda**



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jan 3 2023 a net ionic equation is a chemical equation that shows only the species participating in a chemical reaction and omits the spectator ions spectator ions are ions that

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a net ionic equation is the most accurate representation of the actual chemical process that occurs writing these equations

requires a familiarity with solubility rules acid base reactivity

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apr 25 2017 this is understood when you consider the two chemicals appear unchanged before and after the reaction

therefore the net ionic equation is  $2\text{Cl}^- + 2\text{Ag}^+$

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ions that remain essentially unchanged during a reaction are called spectator ions this means you can ignore them when you write the ionic equation you only need to model how the solid

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may 11 2017 in the net ionic equation any ions that do not participate in the reaction called spectator ions are excluded as a result the net ionic equation shows only the species that

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$1\text{K}_3\text{PO}_4 + \text{Al}(\text{NO}_3)_3 \rightarrow \text{AlPO}_4 + 3\text{KNO}_3$  ionic  $3\text{PO}_4^{3-} + \text{Al}^{3+} \rightarrow \text{AlPO}_4 + 3\text{K}^+$  net ionic  $\text{PO}_4^{3-} + \text{Al}^{3+} \rightarrow \text{AlPO}_4$  s 2 2 fec13 aq 3 mg s

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oct 2 2019 the net ionic equation is a chemical equation for a reaction that lists only those species participating in the reaction the net ionic equation is commonly used in acid base

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jun 23 2022 what remains when the spectator ions are removed is called the net ionic equation which represents the actual chemical change occurring between the ionic

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aug 10 2023 write the full ionic and net ionic equations for this reaction solution the elemental metals magnesium on the reactant side copper on the product side are neutral

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the first step to writing a net ionic equation is balancing the chemical equation present let s use the reaction between sodium chloride and silver nitrate as an example next we write the

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highlights learning objectives by the end of this section you will be able to derive chemical equations from narrative descriptions of chemical reactions write and balance chemical

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the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction notice that in writing the net ionic

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what remains when the spectator ions are removed is called the net ionic equation which represents the actual chemical change occurring between the ionic compounds cl aq

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this net ionic equation indicates that solid silver chloride may be produced from dissolved chloride and silver i ions regardless of the source of these ions these molecular and

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write the overall chemical equation the complete ionic equation and the net ionic equation for the reaction of aqueous silver fluoride with aqueous sodium phosphate to give solid silver

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the complete ionic equation for this reaction is as follows equation 4 13 2ag aq 2f aq 2nh<sub>4</sub> aq cr<sub>2</sub>o<sub>7</sub><sup>2-</sup> aq ag<sub>2</sub>cr<sub>2</sub>o<sub>7</sub> s 2nh<sub>4</sub> aq 2f aq because two nh<sub>4</sub> aq

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