

Assessment

Quiz

Section: Describing Chemical Reactions

In the space provided, write the letter of the term or phrase that best answers the question.

1. The formation of a gas is evidence of
 - a. a chemical change.
 - b. a physical change.
 - c. no change in energy.
 - d. Both (a) and (c).

2. Which of the following may be evidence of a chemical change?
 - a. release of energy as heat
 - b. production of sound
 - c. reduction in temperature
 - d. All of the above

3. In the word equation, iron + copper(II) sulfate → iron(II) sulfate + copper, a product is
 - a. iron(II) sulfate.
 - b. copper.
 - c. iron.
 - d. Both (a) and (b)

4. In the reaction described by the following equation water + energy → hydrogen gas + oxygen gas, energy
 - a. can be considered a reactant.
 - b. is absorbed.
 - c. is released.
 - d. Both (a) and (b)

5. In the unbalanced formula equation carbon + hydrogen → methane, the formula CH₄ represents
 - a. one atom of methane.
 - b. five atoms of methane.
 - c. one molecule of methane.
 - d. one gram of methane.

Holt Chemistry Chapter Test Answers

M Mosston



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Holt Chemistry File ,1998 This reference is a must for students who need extra help reteaching or extra practice The guide moves students through the same concepts as the text but at a slower pace More descriptive detail along with visual algorithms provides a more structured approach Each chapter closes with a large bank of practice problems Book jacket

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practitioners justified intervention by emphasizing the dangers of pelvic disproportion continually enlarging the definition to gain full control of birth Often conditioned by racial assumptions surgical experimentation promoted common use of anesthesia and a radical increase in caesarean sections and birth became a colder more clinical experience

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