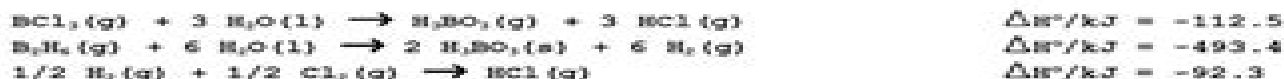


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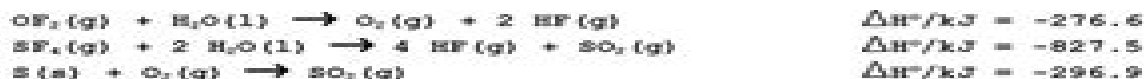
\*\*\*Please check back for the answers. I will repost this document with the answers at a later date\*\*\*

1. Given the following equations and  $\Delta H^\circ$  values, determine the heat of reaction (kJ) at 298 K for the reaction:



**Answer: -1376 kJ**

2. Given the following equations and  $\Delta H^\circ$  values, determine the heat of reaction (kJ) at 298 K for the reaction:



**Answer: -319.5 kJ**

3. Determine  $\Delta H^\circ/\text{kJ}$  for the following reaction using the listed enthalpies of reaction:

# Hess Law Practice Answers

**Patrick E. McMahon, Rosemary  
McMahon, Bohdan Khomtchouk**



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