

## Unit 6a Review questions HL (answers at end)

May 2019 HL, p2 T21

Brass is a copper-containing alloy with many uses. An analysis is carried out to determine the percentage of copper present in three identical samples of brass. The reactions involved in this analysis are shown below.



- (a) (i) Deduce the change in the oxidation numbers of copper and nitrogen in step 1. [2]

Copper:

.....

Nitrogen:

.....

- (ii) Identify the oxidising agent in step 1. [1]

.....

- (b) A student carried out this experiment three times, with three identical small brass nails, and obtained the following results.

Mass of brass = 0.456 g  $\pm$  0.001 g

Titre	1	2	3
Initial volume of 0.100 mol dm <sup>-3</sup> S <sub>2</sub> O <sub>3</sub> <sup>2-</sup> ( $\pm 0.05$ cm <sup>3</sup> )	0.00	0.00	0.00
Final volume of 0.100 mol dm <sup>-3</sup> S <sub>2</sub> O <sub>3</sub> <sup>2-</sup> ( $\pm 0.05$ cm <sup>3</sup> )	28.50	28.60	28.40
Volume added of 0.100 mol dm <sup>-3</sup> S <sub>2</sub> O <sub>3</sub> <sup>2-</sup> ( $\pm 0.10$ cm <sup>3</sup> )	28.50	28.60	28.40
Average volume added of 0.100 mol dm <sup>-3</sup> S <sub>2</sub> O <sub>3</sub> <sup>2-</sup> ( $\pm 0.10$ cm <sup>3</sup> )	28.50		

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