

Gas Laws Worksheet

$$\text{atm} = 760.0 \text{ mm Hg} = 101.3 \text{ kPa} = 760.0 \text{ torr}$$

Boyle's Law Problems: $P_1 V_1 = P_2 V_2$

1. If 22.5 L of nitrogen at 748 mm Hg are compressed to 725 mm Hg at constant temperature. What is the new volume?

$$(748 \text{ mmHg})(22.5 \text{ L}) = (725 \text{ mmHg}) V_2$$

$$V_2 = \frac{(748 \text{ mmHg})(22.5 \text{ L})}{(725 \text{ mmHg})}$$

$$V_2 = 23.2 \text{ L}$$

2. A gas with a volume of 4.0 L at a pressure of 205 kPa is allowed to expand to a volume of 12.0 L. What is the pressure in the container if the temperature remains constant?

$$(4.0 \text{ L})(205 \text{ kPa}) = (12.0 \text{ L}) P_2$$

$$P_2 = \frac{(4.0 \text{ L})(205 \text{ kPa})}{12.0 \text{ L}}$$

$$P_2 = 68.3 \text{ kPa}$$

3. What pressure is required to compress 196.0 liters of air at 1.00 atmosphere into a cylinder whose volume is 26.0 liters?

$$(196.0 \text{ L})(1.00 \text{ atm}) = (26.0 \text{ L}) P_2$$

$$P_2 = \frac{(196.0 \text{ L})(1.00 \text{ atm})}{26.0 \text{ L}}$$

$$P_2 = 7.54 \text{ atm}$$

4. A 40.0 L tank of ammonia has a pressure of 12.7 kPa. Calculate the volume of the ammonia if its pressure is changed to 8.4 kPa while its temperature remains constant.

$$(40.0 \text{ L})(12.7 \text{ kPa}) = (8.4 \text{ kPa}) V_2$$

$$V_2 = \frac{(40.0 \text{ L})(12.7 \text{ kPa})}{8.4 \text{ kPa}}$$

$$V_2 = 60.5 \text{ L}$$

Miscellaneous Gas Law Practice Problems With Answers

**Dr. J. P. Goel, ,D. C. Upadhyay, ,Er.
Meera Goyal**



Miscellaneous Gas Law Practice Problems With Answers:

Physics Class 11 Part I & II combo Scorer Guru Dr. Goyal, ,Upadhyay,Goyal,2023-04-29 VOLUME 1 Mathematical Tools Unit I Physical World and Measurement 1 Physical World 2 Systems of Units and Measurements 3 Significant Figures and Error Analysis 4 Dimensional Analysis Unit II Kinematics 5 Motion in a Straight Line 6 Vector Analysis 7 Motion in a Plane Unit III Laws of Motion 8 Newton s Laws of Motion 9 Friction 10 Uniform Circular Motion Miscellaneous Numerical Examples NCERT Corner Conceptual Problems Exercise Numerical Questions for Practice Multiple Choice Type Questions Unit IV Work Energy and Power 11 Work Energy and Power 12 Centre of Mass 13 Rotational Motion and Moment of Inertia Unit VI Gravitation 14 Gravitation l Log Antilog Table l Value Based Questions VBQ Unit VII Properties of Bulk Matter 16 Pressure of Fluids 17 Viscosity 18 Surface Tension 19 Temperature and Calorimetry 20 Transfer of Heat Unit VIII Thermodynamics 21 First Law of Thermodynamics 22 Second Law of Thermodynamics Unit III Behaviour of Perfect Gases and Kinetic Theory of Gases 23 Behaviour of Perfect Gas and Kinetic Theory Unit IV Oscillations and Waves 24 Oscillations 25 Speed of Mechanical Waves Progressive Waves 26 Superposition of Waves Interference and Beats 27 Reflection of Waves Stationary Waves in Stretched Strings and Organ Pipes 28 Doppler s Effect l Log Antilog Table l Value Based Questions VBQ

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