EARTH AND LIFE SCIENCE 2ND QUARTER G12 ABM- REVIEWER

LESSON 1:

Exogenic Processes

- It is a blanket term for those processes which operate on or close to the surface of the Earth.
- It includes geological phenomena and processes that originate externally to the Earth's surface.

Types of Exogenic Processes

Weathering

- It is the breaking down of rocks into fragments or into smaller size.
- Every rock had undergone weathering, whether it is disintegration (physical weathering) or decomposition (chemical weathering).

Physical Weathering

 Refers to the breakdown of rocks into pieces without any change in its composition.

Types of Physical Weathering

Unloading

- It is the uplifted intrusive igneous rocks.
- It means rocks from beneath the surface go above the surface.
- Weathering are caused by variety of factors such as tectonism and volcanism.

Thermal Expansion & Contractions

- It refers to the effect of temperature to rocks.
- When exposed to high temperature, rocks expand; when exposed to low temperature, rocks contract

Freeze- Thow Weathering

- It is most common in the North pole and South.
- It occurs when a fractured rock absorbs water through its crack then got frozen.
- As mention in Thermal Expansion and contraction, the colder the temperature, the molecule in it expand.

Hydration

- It refers to the amount of water being absorbed by the rock.
- > When hydrated too much or plenty of water is present the rock softens.
- When it is dehydrated too less or lack of waterthe rock hardens.
- > The repeated process of this causes weathering

Salt Crystal Growth

- > It usually happens in the sea or the seaside.
- It happens when salt water comes in contact with rocks.
- The wind can also take part in the salt crystal growth.

Chemical Weathering

Refers to the changes in the composition of rock.

Types of Chemical Weathering

Oxidation

- O* in H*O comes in contact with certain elements in minerals of rocks, and chemical reaction can occur.
- ➤ E.g. Fe and Al Fe² O² and Al² O²

Hydrolysis

- H*O molecules alone or CO* in H*O react with chemical composition/components of rock minerals to create new components or compounds.
- E.g. silicate minerals (containing silica and O*) in igneous rock silicate minerals – clay minerals = KalSi* O* + H*O – HalSi
- > Reactions that produces weak acid.

Solution

- > Rock forming mineral dissolve in water
- E.g. halite mineral containing rock salt (NaCl) susceptible to solution in H*O

Carbonation

- Involves CO* and H*O molecules reacting with, and thereby decomposing rock material.
- ➤ E.g. carbonate rocks (CO²) like limestone (composed of CaCO²) (O² + H²O - H² CO² (carbonic acid) CaCO (limestone) + H² CO² (carbonic acid) + Ca HCO²) calcium bicarbonate

Mass Wasting

- > also called mass movement
- downslope transport of surface materials in direct response to gravity.
- heavier objects have greater downward materials in direct response gravity.
- the force of gravity encourages rock sediments and soil to move downhill or sloping surfaces.

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