

CIE AS LEVEL- CHEMISTRY [9701]

PRACTICAL NOTES

1 ERRORS

$$\text{Estimated Error} = \frac{1}{2} \text{ Smallest Division}$$

$$\% \text{ Uncertainty} = \frac{\text{Estimated Error}}{\text{Reading}}$$

2 TITRATIONS

- Accuracy of Burette = 0.05cm^3
- Always write burette reading to 2dp
- A titre will have no error even though readings have errors if the errors in reading are identical and in the same direction so they will cancel out.
- Two best titres must be within 0.1cm^3 of each other
- If first two titres within 0.1cm^3 then no need for 3rd titre

Use of a Burette	
Advantage	Disadvantage
• Lower % error • More accurately calibrated	• Takes longer to add the FA 2

3 TEMPERATURE

- Record to nearest 0.5°C when thermometer calibrated in 1°C intervals
- Record to nearest 0.1°C when thermometer calibrated in 0.2°C intervals

4 CONVERSIONS

$$100\text{cm} = 10\text{dm} = 1\text{m}$$

$$0^\circ\text{C} = 273^\circ\text{K}$$

$$1\text{cm}^3 \text{ of water} = 1\text{g}$$

5 GRAPHS AND TABLES

- When finding gradient, always use triangle with hypotenuse greater than half of the line
- Label axis with quantity and unit
- Plot graph with fine cross or encircle dots
- For each heading in a table, write the quantity measured with the unit separated with a solidus
- Keep significant figures consistent in values in a table
- Make **only one** table of result for each question

6 PRACTICAL SKILLS

6.1 Measuring a Quantity

Temperature	Use a thermocouple
Volume	Use burette If 25cm^3 use pipette
Mass	Use electronic scale

- Repeat and average values

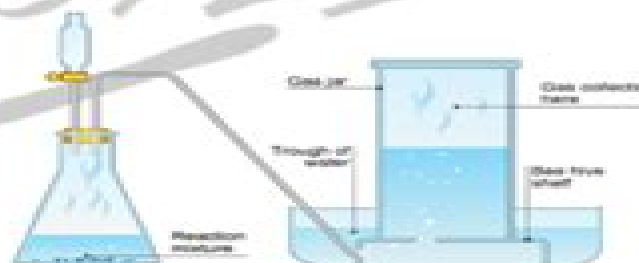
6.2 When Pouring or Adding Substances

- Use a deeper or larger container to hold the substances
- When to use an intermediate apparatus to transfer:
 - If liquid, don't since some may remain in intermediate
 - If solid, make sure all substance has been transferred

6.3 Thermal Experiments

- Insulate container to stop thermal conduction
- Use a lid to seal container to stop thermal convection
- When heating a hydrated salt, heat to constant mass
- Sources of errors in measuring temperature:
 - Heat loss (to the surroundings)
 - Thermometer graduated at 1°C intervals
 - Drying of cup/thermometer
 - Initial temps of both solutions should be taken

6.4 How to Collect CO_2



- Water vapor condenses in water trough

7 SALT ANALYSIS

- If acid added to a salt and produces effervescence, carbonate ion present and write "effervescence produced turns limewater milky"

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Jicheng Xie



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