

1.	(a)	(i)	electrons ...16.... protons .....16 .... neutrons ...16.... <b>All correct (1)</b>	1	[8]
		(ii)	they are isotopes (1) The word 'isotope' or some derivative is essential	1	
		(iii)	S <sup>2-</sup> / sulphide(2-) <b>ion</b> / sulphur -2 ion / sulphur with two extra electrons (1) but <b>not</b> just S or sulphur	1	
	(b)	(i)	X(g) + e <sup>-</sup> → X <sup>-</sup> (g) (S(g) + e <sup>-</sup> → S <sup>-</sup> (g)) (1) X <sup>-</sup> (g) + e <sup>-</sup> → X <sup>2-</sup> (g) (S <sup>-</sup> (g) + e <sup>-</sup> → S <sup>2-</sup> (g)) (1) can have <b>Z</b> in place of <b>X<sup>2-</sup></b> but not <b>Z<sup>2-</sup></b> Penalize omission of state symbol (g) only once.	2	
		(ii)	adding a negative onto a <u>negative species</u> requires energy to overcome repulsion (1) <b>or</b> second electron repelled by negative charge <b>Not</b> second electron repelled by first	1	
		(c)	(Bdecay from) Y produces only one isotope( <sup>39</sup> Cl) (1) natural sample contains two isotopes (35 and 37) (1)	2	
	2.	(i)	magnesium / lead / iron (1)	1	
		(ii)	iron (1)	1	
		(iii)	carbon / phosphorus (1)	1	
		(iv)	magnesium / lead (1)	1	
		(v)	carbon / lead (1)	1	
	(b)		Helium (1) because it is the <b>smallest</b> atom (1) with an unshielded nucleus (1) Hydrogen can score max 1 for 'unshielded nucleus' If candidate chooses Carbon, i.e. uses list from part (a), can score 1 mark for reference to 'smallest atom'	3	
		(c)	(i) ability of a positive ion / cation to attract electrons / distort / pull the electron cloud (1) around an anion (1)	2	
	(ii)		size (1) <b>Not</b> atomic radius charge (1) charge density <b>worth</b> (2) volume of ion can score in place of size	2	
		(iii)	magnesium chloride magnesium ion smaller than calcium ion (1) polarizes / distorts / attracts electrons around chloride more (1)	2	
					[14]

# June 2013 Chemistry Unit 5 Mark Scheme

**Yicheng Fang**



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