

3. A student wants to remove an acid impurity from an organic liquid.  
What should the student do?

A. Add  $\text{Na}_2\text{CO}_3(\text{aq})$   
B. Reflux the mixture  
C. Add  $\text{Br}_2$   
D. Add  $\text{MgSO}_4$

Your answer ☐

[1]

4. For which purpose is distillation used?
- A. to allow a liquid to boil without the loss of vapour  
B. to purify a liquid product  
C. to remove an involatile impurity  
D. to allow further reaction without the loss of product

Your answer ☐

[1]

5. The reaction below produces a chloroalkane.



Which row shows the steps to purify the liquid product in the correct order?

A	Use a separating funnel	Distil	Dry	Remove unreacted $\text{HCl}$
B	Remove unreacted $\text{HCl}$	Use a separating funnel	Dry	Distil
C	Remove unreacted $\text{HCl}$	Use a separating funnel	Distil	Dry
D	Use a separating funnel	Remove unreacted $\text{HCl}$	Distil	Dry

Your answer ☐

[1]

# June 2013 Chemistry Ocr Mark Scheme

**Judd E. Hollander**



**June 2013 Chemistry Ocr Mark Scheme:**

## The Enigmatic Realm of **June 2013 Chemistry Ocr Mark Scheme**: Unleashing the Language is Inner Magic

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