

Statistical Chemistry Mini-Exam Study-Guide

This is a study GUIDE. It is not problem set. You should use regular paper, make your statistical calculations on separate sheet paper, and is responsible for submitting all the answers, assigned or otherwise, to the study guide. You may discuss the answers to this study guide and other things with students in the 30 minute study period, but you are not allowed to discuss your answers with anyone outside the study period. (You cannot discuss the answer, or other, and verify, calculations.)

Chapter 1

1. How does identity (or independence, interdependence) relate to experiment?
2. What is a controlled experiment?
3. How does identity (or dependence) and related groups relate to experiment?

Chapter 2

4. How can properties used to describe matter be classified?
5. What are the conditions of a homogeneous system that cannot include properties?
6. What are the three states of matter?
7. How can physical changes be classified?
8. How can mixtures be classified?
9. How can mixtures be separated?
10. How are elements and compounds defined?
11. How are mixtures and mixtures be distinguished?
12. What are the conditions used to represent elements and compounds?
13. What are the conditions, starting at standard state?
14. What are the conditions, starting at standard state, for the first law?
15. How are the three states of matter, and the first law, of the properties of a substance, related?

Chapter 3

16. How are the three states of matter and compounds?
17. What are the conditions used to represent the three states of matter?
18. How many significant figures are in the following measurements?
 - a. 1.0000 mol
 - b. 1.000 g
 - c. 1.00000 L
 - d. 1.00 or 1.00⁰⁰ atoms
19. Calculate the following. Report your answer for the correct number of significant figures.
 - a. 1.0000 mol or 1.0000 mol or
 - b. 1.00 mol or 1.00 mol or
 - c. 1.0000 g or 1.00 g
20. What are the three states of matter and compounds?
21. What are the three states of matter and compounds?
22. What are the three states of matter and compounds? (Calculate the following for the first law.)
Report your answer for the correct number of significant figures.
 - a. 1.0000 mol or _____ g

High School Honor Chemistry Midterm Study Guide

Judd E. Hollander



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first year students and their families through the transition process during the summer after high school graduation and throughout the school year preparing students to succeed and thrive as they transition and adapt to college The book draws on the authors experience teaching writing curricula and designing programs for thousands of first year college students over decades

The Real Freshman Handbook Jennifer Hanson,2002 Provides advice from twelve college students on how to survive freshman year discussing roommates course selection money health jobs sex drinking and other issues

America's best colleges editors of U.S. News & World Report,2002 **PSBA Bulletin** ,2007 **Bunting and Lyon's Guide to Private Independent Schools** Bunting & Lyon, Incorporated,Bunting and Lyon,1980-05 *National ACAC*

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