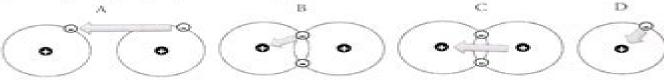
#### Read This!

Electronegativity is a measure of the ability of an atom's nucleus to attract electrons from a different atom within a covalent bond. A higher electronegativity value correlates to a stronger pull on the electrons in a bond. This value is only theoretical. It cannot be directly measured in the lab.

 Using the definition stated in the Rend Thic box above, select the best visual representation for electronegativity. Explain your reasoning.



- 13. Locate the electronegativity values in Model 1.
  - at. What is the trend in electronegativity going down a group in Model 1?

Electronegativity decreases as you go down a group (vertical column).

 Explain the existence of the trend described in part a in terms of atomic structure and Coulombic attraction.

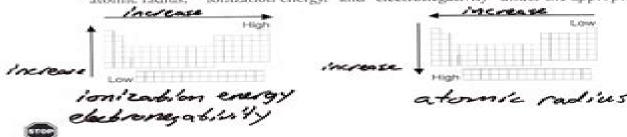
The ability to attract electrons is based upon attractive force.
The granter the distance between the nucleus and valence electrons the less the attractive force. Less force means a lower electron egativity. What is the trend in electron egativity going across a period in Model 1?

Electrone alice the imposer from left to what a man a manual throughout.

Electronegativity increases from left to right across a period (hon'ron to)

The greater the number protons in an above, in a particular period, the greater the altractive force between the nucleus a valence electrons. The greater attractive force means a higher electronegativity.

4. The two diagrams below can summarize each of the three trends discussed in this activity. Write "atomic radius," "ionization energy," and "electronegativity" under the appropriate diagram.



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Innovative Methods of Teaching and Learning Chemistry in Higher Education Ingo Eilks, Bill Byers, 2015-11-06
Two recent initiatives from the EU namely the Bologna Process and the Lisbon Agenda are likely to have a major influence on European Higher Education It seems unlikely that traditional teaching approaches which supported the elitist system of the past will promote the mobility widened participation and culture of life long learning that will provide the foundations for a future knowledge based economy There is therefore a clear need to seek new approaches to support the changes which will inevitably occur The European Chemistry Thematic Network ECTN is a network of some 160 university chemistry departments from throughout the EU as well as a number of National Chemical Societies including the RSC which provides a discussion forum for all aspects of higher education in chemistry This handbook is a result of one of their working groups who identified and collated good practice with respect to innovative methods in Higher Level Chemistry Education It provides a comprehensive overview of innovations in university chemistry teaching from a broad European perspective The generation of this book through a European Network with major national chemical societies and a large number of chemistry departments as members make the book unique The wide variety of scholars who have contributed to the book make it interesting and invaluable reading for both new and experienced chemistry lecturers throughout the EU and beyond The book is aimed at chemistry education at universities and other higher level institutions and at all academic staff and anyone

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