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Arshad Iqbal



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web we utilize the equation that relate amount remaining initial mass and number of half lives $n = \frac{t}{T_{1/2}} \times \log_2 \left(\frac{m_0}{m} \right)$
125 g 5 what is the half life of an isotope that is 80 remained after 16 days solution remaining 80 100 therefore $n = \frac{t}{T_{1/2}} \times \log_2 \left(\frac{m_0}{m} \right)$
now using the half life equation in example 1 we have

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web section learning objectives by the end of this section you will be able to do the following explain radioactive half life and its role in radiometric dating calculate radioactive half life and solve problems associated with radiometric dating section key terms half life and the rate of radioactive decay unstable nuclei decay

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web solution 7.2.4.3 half lives 1.2.0.125 the amount remaining after 3 half lives 100.0 g $\times$ 0.125 = 12.5 g remaining problem 2
pd 100 has a half life of 3.6 days if one had 6.02×10^{23} atoms at the start how many atoms would be present after 20.0 days

solution 20.0 $\div$ 3.6 = 5.56 half lives

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