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LABORATORY

Preparing and Diluting Solutions Student Laboratory Kit

Introduction

Solutions are homogeneous mixtures. The most common solutions are aqueous solutions, which are solutions in which water is the solvent.

Concepts

- 1. Solvent
- 2. Solute

Background

Preparing Solutions

The amount of solute that is dissolved in a solution is called the concentration. The concentration of a solution can be expressed in many ways. The most common way is to express it as a percentage. For example, a 10% solution means that there is 10 grams of solute in 100 grams of solution.

When you prepare a solution, you must know the amount of solute and the amount of solvent. The amount of solute is usually given in grams, and the amount of solvent is usually given in milliliters.

Remember, always use the correct units when you prepare a solution. If you use the wrong units, you will not get the correct concentration.

When you dilute a solution, you are adding more solvent to it. This decreases the concentration of the solution. For example, if you have a 10% solution and you add more solvent, the concentration will be less than 10%.

Remember, always use the correct units when you dilute a solution. If you use the wrong units, you will not get the correct concentration.



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