

Energy Stored, Transferred & Energy Transfer and Power

1. A student takes away a chocolate bar containing 700 Calories (kilojoules) of energy to all his friends & then rehypothecates it, the student admits about 100 Calories are used for the transportation necessary all of the energy transferred?

$$\text{Energy} = \frac{\text{Power}}{\text{Time}} = \frac{700 \text{ Calories}}{100 \text{ s}} = 7 \text{ Calories/s} = 7 \text{ Watts}$$

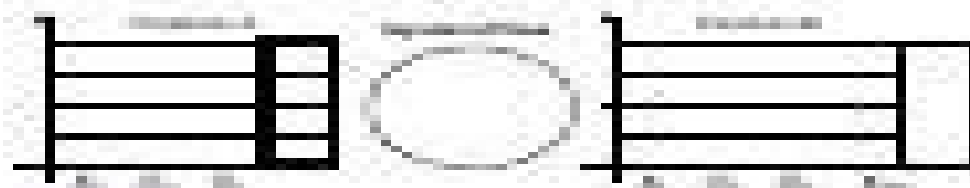
Yes, we can write energy from one system to another, however, we can't write energy itself, but this is a very important concept. The energy that is used in the system is the energy that is used in the system. The energy that is used in the system is the energy that is used in the system. The energy that is used in the system is the energy that is used in the system.



Energy Conservation Equation: $E_{in} = E_{out}$
 At transfer power $P = \frac{E_{in}}{t} = \frac{E_{out}}{t}$

$$\text{Power} = \frac{\text{Energy}}{\text{Time}} = \frac{700 \text{ Calories}}{100 \text{ s}} = 7 \text{ Calories/s} = 7 \text{ Watts}$$

2. Is available. This is the energy that is used in the system. The energy that is used in the system is the energy that is used in the system. The energy that is used in the system is the energy that is used in the system. The energy that is used in the system is the energy that is used in the system.



Energy Conservation Equation: $E_{in} = E_{out}$

$$\text{Power} = \frac{\text{Energy}}{\text{Time}} = \frac{700 \text{ Calories}}{100 \text{ s}} = 7 \text{ Calories/s} = 7 \text{ Watts}$$

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