

Unit 5 – Chemical Reactions

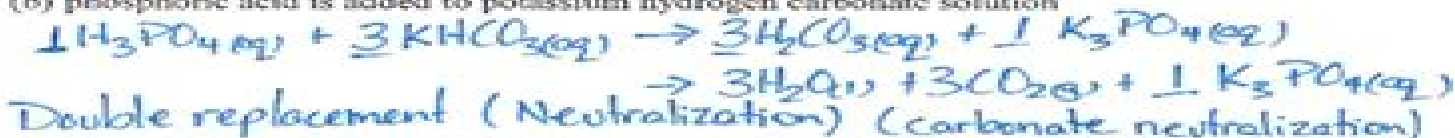
13 Write complete balanced chemical equations for the following. Classify each reaction as combination, decomposition, single replacement, double replacement, neutralization, or combustion. Don't forget your subscripts!

(a) potassium reacts with bromine



Combination

(b) phosphoric acid is added to potassium hydrogen carbonate solution



Double replacement (Neutralization) (carbonate neutralization)

(c) magnesium is added to nickel (II) nitrate solution



Single Replacement.

(d) hydrochloric acid is added to ammonium hydroxide.



Double Replacement (neutralization)

(e) nickel is added to magnesium chloride solution



(f) copper (II) sulphate pentahydrate is produced in a combination reaction



Combination

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K Morrison



Modeling Chemistry Unit 5 Test B V2 0 Key:

Modeling Chemistry Unit 5 Test B V2 0 Key Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Modeling Chemistry Unit 5 Test B V2 0 Key**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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