Name	
Date	Per

#### **Mixed Practice: Names and Formulas**

First determine whether each of the following compounds is *ionic* (M + NM), polyatomic (3 or more atoms), or covalent (2 NM's). Then write the name for each compound.

1) NO <sub>2</sub>		
2) NaBr		
3) SiO <sub>2</sub>		
4) P <sub>2</sub> Br <sub>4</sub>		
5) FeSO <sub>4</sub>		
6) SF <sub>6</sub>		
7) Li <sub>2</sub> S		
8) MgBr <sub>2</sub>		
9) N <sub>2</sub> S		
10) Be(OH) <sub>2</sub>		
11) SO <sub>3</sub>		
12) Cu <sub>2</sub> S		
13) BF <sub>3</sub>		
14) I <sub>2</sub>		
15) Pb <sub>3</sub> (PO <sub>3</sub> ) <sub>2</sub>		

# **Modeling Chemistry Unit 4 Answers**

JL Elias

**Modeling Chemistry Unit 4 Answers:** 

Unveiling the Magic of Words: A Review of "Modeling Chemistry Unit 4 Answers"

In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "Modeling Chemistry Unit 4 Answers," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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