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#### **Mixed Practice: Names and Formulas**

First determine whether each of the following compounds is *ionic* (M + NM), polyatomic (3 or more atoms), or covalent (2 NM's). Then write the name for each compound.

1)	NO <sub>2</sub>	
2)	NaBr	_
3)	SiO <sub>2</sub>	
4)	P <sub>2</sub> Br <sub>4</sub>	
5)	FeSO <sub>4</sub>	
6)	SF <sub>6</sub>	

- 7) Li<sub>2</sub>S \_\_\_\_\_
- 8) MgBr<sub>2</sub> \_\_\_\_\_
- 9) N<sub>2</sub>S \_\_\_\_\_
- 10) Be(OH)<sub>2</sub>
- 11) SO<sub>3</sub>\_\_\_\_\_
- 12) Cu<sub>2</sub>S \_\_\_\_\_
- 13) BF<sub>3</sub>\_\_\_\_\_
- 14) I<sub>2</sub>\_\_\_\_\_
- 15) Pb<sub>3</sub>(PO<sub>3</sub>)<sub>2</sub>\_\_\_\_\_

# **Modeling Chemistry Unit 4 2 Answers**

Jianjun Gao

**Modeling Chemistry Unit 4 2 Answers:** 

#### Modeling Chemistry Unit 4 2 Answers Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has be much more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Modeling Chemistry Unit 4 2 Answers**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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