

**STATE OF NURSING (PROSECUTION, PILAPIL'S GROUP) VS BINNIE
(DEFENDANT, CAMARILLO'S GROUP)**

Members of Defendant's Side:

Bailiff- Gular, Nina
Defense Witness 1- Hermosisima, Mary Noelle (Binnie)
Defense Witness 2 - Gomonit ,Christian (Jeremy)
Defense Witness 3 - Garcia, Francheska (Gynecologist)
Defense Attorney 1 - Carampil, Andrea
Defense Attorney 2 - Camarillo, Andre
Notary - Reyes,Cherry
Jury 1 - Palatan, Christhel
Jury 2 - Padilla, Shanley
Jury 3 - Silvestre, Chantelle
Jury 4 - Valladares, Kristyl

Members of Persecution Side:

Prosecution attorney 1 - Pilapil, Bradley
Prosecution attorney 2 - Palomar, Denzel
Prosecution Witness 1 - Delima, Cedrix (Alberto)
Prosecution Witness 2 - Ilagan, Marie Alvi (Janice)
Prosecution Witness 3 - De Reyes, Sark (Magda Lena)
Jury 1 - Trinidad, Shaine
Jury 2 - Lampong, Jehann
Jury 3 - Sagaral, Marielle
Jury 4 - Pagulong, Yvonne
Judge - Pata, Gabriel

Notary (REYES, CHERRY): The above named defendant is accused of

Count 1: Murder in the First Degree

Committed as follows

The Defendant is charged for the murder of her newborn child

Bailiff (GULAR): Please Rise. The court of the second judicial circuit, criminal division is now in session, the Honorable Judge Ms. BLANK Presiding

Judge (PATA): Everyone but the jury may be seated. Mr. Camarillo please swear in the Jury

Mock Trial Scripts Nursing Malpractice Cases

Christian Drosten



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 $2 \text{ CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g}) \rightarrow \text{C}_2\text{H}_2(\text{g}) + \frac{5}{2} \text{O}_2(\text{g}) \quad \Delta h = 94.5 \text{ kJ}$
 $\text{H}_2\text{O}(\text{g}) \rightarrow \text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \quad \Delta h = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{g}) \quad \Delta h = 283 \text{ kJ}$

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 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 2 \text{H}_2\text{O}(\text{l}) \quad \Delta h = 1411 \text{ kJ/mole}$
 $\text{C}_2\text{H}_6(\text{g}) + 7/2 \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{l}) \quad \Delta h = 1560 \text{ kJ/mole}$
 $\text{H}_2(\text{g}) + 1/2 \text{O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l}) \quad \Delta h = 285 \text{ kJ/mole}$

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